



The ternary system cerium–rhodium–silicon

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ABSTRACT

Phase relations have been established in the ternary system Ce–Rh–Si for the isothermal section at 800 °C based on X-ray powder diffraction and EPMA on about 80 alloys, which were prepared by arc melting under argon or by powder reaction sintering. From the 25 ternary compounds observed at 800 °C 13 phases have been reported earlier. Based on XPD Rietveld refinements the crystal structures for 9 new ternary phases were assigned to known structure types. Structural chemistry of these compounds follows the characteristics already outlined for their prototype structures: τ_7 —Ce₃RhSi₃ (Ba₃Al₂Ge₂-type), τ_8 —Ce₂Rh_{3-x}Si_{3+x} (Ce₂Rh_{1.35}Ge_{4.65}-type), τ_{10} —Ce₃Rh_{4-x}Si_{4+x} (U₃Ni₄Si₄-type), τ_{11} —CeRh₆Si₄ (LiCo₆P₄-type), τ_{13} —Ce₆Rh₃₀Si_{19.3} (U₆Co₃₀Si₁₉-type), τ_{18} —Ce₄Rh₄Si₃ (Sm₄Pd₄Si₃-type), τ_{21} —CeRh₂Si (CeIr₂Si-type), τ_{22} —Ce₂Rh_{3+x}Si_{1-x} (Y₂Rh₃Ge-type) and τ_{24} —Ce₈(Rh_{1-x}Si_x)₂₄Si (Ce₈Pd₂₄Sb-type). For τ_{25} —Ce₄(Rh_{1-x}Si_x)₁₂Si a novel bcc structure was proposed from Rietveld analysis. Detailed crystal structure data were derived for τ_3 —CeRhSi₂ (CeNiSi₂-type) and τ_6 —Ce₂Rh₃Si₅ (U₂Co₃Si₅-type) by X-ray single crystal experiments, confirming the structure types. The crystal structures of τ_4 —Ce₂₂Rh₂₂Si₅₆, τ_5 —Ce₂₀Rh₂₇Si₅₃ and τ_{23} —Ce_{33.3}Rh_{58.2-55.2}Si_{8.5-11.5} are unknown. High temperature compounds with compositions Ce₁₀Rh₅₁Si₃₃ (U₁₀Co₅₁Si₃₃-type) and CeRhSi (LaIrSi-type) have been observed in as-cast alloys but these phases do not participate in the phase equilibria at 800 °C.

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1. Introduction

Six ternary compounds from the title system, CeRh₂Si₂ (ThCr₂Si₂-type) [1], CeRhSi₃ (BaNiSn₃-type) [2,3], CeRhSi₂ (CeNiSi₂-type), CeRh₃Si₂ (CeCo₃B₂-type), Ce₂Rh₃Si₅ (U₂Co₃Si₅-type) [4] and Ce₂RhSi₃ (Er₂RhSi₃-type) [5] were described prior to the first investigation of the entire Ce–Rh–Si phase diagram at 600 °C by Shapiev [6]. In his work 17 ternary compounds were found, however, only the approximate composition for 11 new compounds was derived from phase analysis on X-ray powder photographs: Ce₁₀Rh₇₅Si₁₅, Ce₁₀Rh₆₀Si₃₀, Ce₁₀Rh₅₅Si₃₅, Ce_{27.5}Rh_{43.5}Si₂₉, Ce_{33.3}Rh_{56.3}Si₁₀, Ce_{33.3}Rh_{46.7}Si₂₀, Ce_{33.3}Rh_{33.3}Si_{33.4}, Ce_{33.3}Rh_{26.3}Si₄₀, Ce₄₀Rh₁₅Si₄₅, Ce₄₅Rh₂₅Si₃₀, Ce_{51.5}Rh₄₀Si_{8.5}. Since 1993 additional knowledge has been acquired and crystal structures of several ternary phases were determined, i.e. Ce₆Rh₃₀Si₁₉ (Sc₆Co₃₀Si₁₉-type) [7], Ce₂Rh₁₂Si₇ (Ho₂Rh₁₂As₇-type) [7], Ce₂Rh₁₅Si₇ (own-type) [8], Ce₃Rh₃Si₂ (own-type) [9], CeRh₂Si

(inverse CeNiSi₂-type) [10], Ce₃Rh₂Si₂ (La₃Ni₂Ga₂-type) [11]. More accurate version of the crystal structure was established for CeRh₃Si₂ (ErRh₃Si₂-type—an orthorhombic superstructure of CeCo₃B₂ [12]) and for Ce₂RhSi₃ (Er₂RhSi₃-type) a higher symmetry from *P62c* to *P63/mmc* was proposed [13]. Physical properties of several compounds from the Ce–Rh–Si system were investigated and revealed interesting electrical and/or magnetic anomalies, such as heavy-fermion behavior (including non-centrosymmetric heavy-fermion superconductivity in CeRhSi₃—see [14] and for latest investigations [15–20]) and/or Kondo-lattices as well as valence instabilities, e.g. for CeRh₂Si₂ [21–33], CeRhSi₂ [34–36], Ce₂RhSi₃ [5,37–42], CeRh₂Si [43,44], Ce₂Rh₃Si₅ [45,46], CeRh₃Si₂ [47–49], Ce₃Rh₃Si₂ [50] and Ce₂Rh₂Si₅ [51] (for the latter phase no crystal structure data were given). Physical properties of four compounds with known crystal structure, i.e. Ce₂Rh₁₅Si₇, Ce₂Rh₁₂Si₇, Ce₆Rh₃₀Si₁₉ and Ce₃Rh₂Si₂, have not been reported yet.

As the early data presented by Shapiev [6] do not fully comply with information hitherto given, we reinvestigated the phase equilibria in the Ce–Rh–Si system. In the present paper we provide comprehensive knowledge on the phase equilibria at 800 °C and crystal structures of intermetallics. The annealing temperature of 800 °C was chosen in order to achieve enhanced

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Table 1
Crystallographic data of solid phases in the Ce–Rh–Si system.

Phase	Space group	Lattice parameters (nm)			Comments
		<i>a</i>	<i>b</i>	<i>c</i>	
(δ Ce)	$Im\bar{3}m$	0.412			[67]
798–726 [67]	<i>W</i>				
(γ Ce)	$Fm\bar{3}m$	0.51610			[67]
726–61 [67]	<i>Cu</i>				
(β Ce)	$P6_3/mmc$	0.36810		1.1857	[67]
61–(–177) [67]	αLa				
(α Ce)	$Fm\bar{3}m$	0.485			[67]
≤ -177 [67]	<i>Cu</i>				
(Rh)	$Fm\bar{3}m$	0.38032			[67]
≤ 1963 [67]	<i>Cu</i>				
(Si)	$Fm\bar{3}m$	0.54306			[67]
≤ 1414 [67]	<i>C (diamond)</i>				
Ce ₅ Si ₃	$I4/mcm$	0.7868		1.373	[68]
≤ 1260 [59]	Cr ₅ B ₃	0.7878		1.367	[59]
Ce ₅ (Si _{1–x} Rh _x) ₃		0.7873		1.3796	0 $\leq x \leq 0.3$ [69] at 600 °C
		0.78592		1.3657	$x_{max}=0.3$ [69] at 600 °C
Ce ₃ Si ₂	$P4/mbm$	0.7780		0.4367	[59]
≤ 1335 [59]	U ₃ Si ₂	0.77908(4)		0.43690(3)	^d
Ce ₅ Si ₄	$P4_12_12$	0.7936		1.5029	[59]
≤ 1500 [59]	Zr ₅ Si ₄	0.79438(3)		1.5098(1)	^d
CeSi	$Pnma$	0.8298	0.3961	0.5959	[59]
≤ 1630 [59]	FeB	0.82832(3)	0.39756(1)	0.59601(2)	^d
CeSi _{1.34}	$Cmcm V_2B_3$	0.44035	2.48389	0.39517	[60] at 35 K
	(Nd ₂ Si _{3–x})				
CeSi _{1.67}	$Imma$	0.4109	0.4189	1.3917	Si-saturated (sat.) [59]
≤ 1725 [59]	GdSi _{2–x}	0.4190	0.4113	1.3906	Rh-sat. [59]
		0.41220(1)	0.41913(1)	1.3914(1)	^d
CeSi _{2–x}	$I4_1/amd$	0.4192		1.3913	$x=0$ [59]
≤ 1575 [59]	ThSi ₂	0.41897(3)		1.3932(1)	$x=0^d$
Ce(Si _{1–y} Rh _y) _{2–x}		0.41907(1)		1.3937(1)	$x=0, y=0.06^d$
		0.41845(1)		1.4025(1)	$x=0, y=0.105^d$
		0.41781(1)		1.4138(1)	$x=0, y_{max}=0.144^d$
Ce ₇ Rh ₃	$P6_3mc$	1.0051		0.6378	[61]
≤ 755 [61]	Th ₇ Fe ₃	1.0005		0.6356	[70]
		1.0023		0.6376	[71]
Ce ₅ Rh ₃	$P4/ncc$	1.1260		0.6450	[61]
≤ 725 [61]	Pu ₅ Rh ₃				
Ce ₃ Rh ₂	$R\bar{3}$	0.8858		1.6794	[61]
≤ 815 [61]	Er ₃ Ni ₂	0.8835		1.676	[72]
Ce ₅ Rh ₄	$Pnma$	0.7467	1.4841	0.7624	[61]
≤ 955 [61]	Sm ₅ Ge ₄	0.7434	1.486	0.7604	[73]
		0.7486(1)	1.4788(2)	0.7657(1)	^d
CeRh	$Cmcm$	0.3845	1.0964	0.4174	[61]
≤ 1055 [61]	CrB	0.3852	1.0986	0.4152	[74]
		0.3855	1.0966	0.4153	[75]
		0.3844(1)	1.0946(2)	0.4174(1)	^d
CeRh ₂	$Fd\bar{3}m$	0.7545			Alloy at 65 at% Rh [61]
≤ 1450 [61]	MgCu ₂	0.7543			Alloy at 70 at% Rh [61]
		0.7538			[76]
		0.7547			Ce-sat. [76]
		0.7534			Rh-sat. [76]
		0.7550			[77]
		0.7539			[78]
		0.75387(2)			Rh-sat. ^d
Ce(Rh _{1–x} Si _x) ₂		0.75256(2)			0 $\leq x \leq 0.10^d$
		0.75220(5)			$x=0.04^d$
CeRh ₃	$Pm\bar{3}m$	0.4022			$x_{max}=0.10^d$
≤ 1550 [61]	Cu ₃ Au	0.4023			Alloy at 70 at% Rh [61]
		0.4012			[79]
		0.4020			[70]
		0.4024			[76]
					[80]
CeRh ₃ Si _{1–x}		0.40455(1)			0.81 $\leq x \leq 1.00^d$
		0.40616(1)			$x=0.91^d$
		0.40912(1)			$x=0.87^d$
					$x_{min}=0.81^d$
Rh ₂ Si	$Pnma$	0.5404	0.3930	0.7390	[81]
≤ 1622 [62]	αCo_2Si	0.54109(1)	0.39294(1)	0.73855(1)	^d
Rh ₅ Si ₃	$Pbam$	0.5322	1.0126	0.3897	[81]
≤ 1470 [62]	Rh ₅ Ge ₃	0.53181(1)	1.01257(2)	0.38966(1)	^d
Rh ₂₀ Si ₁₃	$P6_3/m$	1.1851		0.3623	[82]
1050 – 1225 [62]	Rh ₂₀ Si ₁₃	1.1853(1)		0.36128(1)	^d

Table 1 (continued)

Phase	Space group	Lattice parameters (nm)			Comments
		<i>a</i>	<i>b</i>	<i>c</i>	
Rh ₃ Si ₂ metastable [62]	<i>P</i> $\bar{3}$ <i>m</i> 1	0.3965		0.5051	[81]
α RhSi ≤ 1030 [62] Si-sat. ≤ 1080 [62] Rh-sat.	Rh ₃ Si ₂ <i>Pnma</i> MnP	0.5545 0.555 0.55466(5) 0.55517(3)	0.3067 0.3084 0.30707(2) 0.30707(1)	0.6376 0.6350 0.63658(5) 0.63736(3)	Rh-sat. 49.8 at% Si [81] Si-sat. 50.7 at% Si [81] Si-sat. 50 at% Rh ^d Rh-sat. 50.8 at% Rh ^d Rh-sat. 49.8 at% Si [81] Si-sat. 50.7 at% Si [81] Rh-sat. ^d as-cast
β RhSi 1030–1452 [62] Si-sat. 1080–1452 [62] Rh-sat.	<i>P</i> 2 ₁ 3 FeSi	0.4685 0.4674 0.46822(2)			
Rh ₄ Si ₅ ≤ 1030 [62]	<i>P</i> 12 ₁ / <i>m</i> Rh ₄ Si ₅	1.234 1.2335(1)	0.3512 $\beta=100.18^\circ$ 0.35071(2) $\beta=100.18(1)^\circ$	0.5929 0.59285(4)	[81] ^d
Rh ₃ Si ₄ ≤ 1040 [62] τ_1 , CeRhSi ₃	<i>Pnma</i> Rh ₃ Si ₄ <i>I</i> 4 <i>mm</i> BaNiSn ₃	1.880 1.8800(1) 0.4204 0.4269 0.4240 0.4235 0.4244 0.4237 0.42311(1)	0.3613 0.36048(2)	0.5808 0.58089(3) 0.974 0.9738 0.9801 0.9788 0.9813 0.9760 0.97816(3)	[81] ^d [2,3] [34] [83] [84] [85] [17] [20] ^d
τ_2 , Ce ₂ RhSi ₃	<i>P</i> 6 ₃ / <i>mmc</i> Er ₂ RhSi ₃	0.8262 0.8327 0.8240 0.8237		0.8439 0.8516 0.8444 0.8445	[5,13] [37] [39] [40] [41]
Ce ₂ Rh _{1-x} Si _{3+x}		0.82359(2) 0.82255(1) 0.82162(1) 0.82068(1)		0.84212(3) 0.84605(1) 0.84804(2) 0.85030(2)	$0 \leq x \leq 0.24^d$ $x_{\min}=0^d$ $x=0.10^d$ $x=0.18^d$ $x_{\max}=0.24^d$
τ_3 , CeRhSi ₂	<i>Cmcm</i> CeNiSi ₂	0.42661 0.4289 0.4310	1.6758 1.6805 1.6743	0.41708 0.4232 0.4216	[34] [35] [36]
CeRh _{1-x} Si _{2+x}		0.42629(1) 0.42582(1) 0.42604(2) 0.42566(4)	1.67456(3) 1.67526(2) 1.67737(6) 1.6768(2)	0.41731(1) 0.41773(1) 0.41787(1) 0.41763(4)	$0 \leq x \leq 0.32^d$ $x_{\min}=0.0^d$ $x=0.15^d$ $x=0.24^d$ SC $x_{\max}=0.32^d$ ^d
τ_4 , Ce ₂₂ Rh ₂₂ Si ₅₆ Ce ₂ Rh ₂ Si ₅ τ_5 , Ce ₂₀ Rh ₂₇ Si ₅₃ ^a	Unknown <i>t</i> ^{**}	0.41391(1)		0.99650(2)	[51] ^d
τ_6 , Ce ₂ Rh ₃ Si ₅	BaAl ₄ -deriv. <i>Ibam</i> U ₂ Co ₃ Si ₅	0.9874 0.989	1.183 1.187	0.5822 0.582	[86] [46]
τ_7 , Ce ₃ RhSi ₃	<i>I</i> mmm Ba ₃ Al ₂ Ge ₂	0.99118(2) 0.41528(1)	1.17558(3) 0.42548(1)	0.58313(1) 1.81906(5)	^d SC ^d
τ_8 , Ce ₂ Rh _{3-x} Si _{3+x}	<i>P</i> mmm Ce ₂ Rh _{1.35} Ge _{4.65}	0.41035(2) 0.41103(1) 0.41167(2)	0.41091(2) 0.41047(1) 0.40920(2)	1.72633(7) 1.72595(2) 1.72339(6)	$0.05 \leq x \leq 0.36^d$ $x_{\min}=0.05^d$ $x=0.27^d$ $x_{\max}=0.36^d$
τ_9 , CeRh ₂ Si ₂	<i>I</i> 4/ <i>mmm</i> ThCr ₂ Si ₂	0.4086 0.4093 0.4075 0.40840 0.40834 0.40828 0.409 0.4087 0.4088 0.40883(1)		1.017 1.0185 1.013 1.01693 1.01713 1.01705 1.018 1.017 1.0178 1.01748(2)	[1] [21] [23] <i>T</i> =47.5 K [22] <i>T</i> =32 K [22] <i>T</i> =10 K [22] r.t. [22] [87] [47] ^d
τ_{10} , Ce ₃ Rh _{4-x} Si _{4+x}	<i>I</i> mmm U ₃ Ni ₄ Si ₄	0.40676(1) 0.40813(2) 0.69881(2)	0.41409(1) 0.41322(2)	2.44727(4) 2.4465(1) 0.37808(2)	$0.0 \leq x \leq 0.1^d$ $x_{\min}=0^d$ $x_{\max}=0.1^d$ ^d
τ_{11} , CeRh ₆ Si ₄	<i>P</i> $\bar{6}$ <i>m</i> 2 LiCo ₆ P ₄	0.69881(2)		0.37808(2)	
τ_{12} , CeRh ₃ Si ₂	<i>I</i> mma ErRh ₃ Si ₂	0.7128 0.7125 0.71121	0.9725 0.9709 0.96810	0.5595 0.5592 0.55828	[47] [88] SC [48] SC

Table 1 (continued)

Phase	Space group	Lattice parameters (nm)			Comments
		<i>a</i>	<i>b</i>	<i>c</i>	
Temperature range (°C)	Prototype				
τ_{13} , Ce ₆ Rh ₃₀ Si _{19.3}	<i>P6₃/m</i> U ₆ Co ₃₀ Si ₁₉	0.71252(1)	0.97220(2)	0.55932(1)	^d
		2.2300(1)		0.38398(1)	^d
τ_{14} , Ce ₆ Rh ₃₀ Si ₁₉	<i>P6₃/m</i> Sc ₆ Co ₃₀ Si ₁₉	1.5698		0.38571	[7] SC
		1.57257(2)		0.38569(1)	^d
τ_{15} , Ce ₂ Rh ₁₂ Si ₇	<i>P6₃/m</i> Ho ₂ Rh ₁₂ As ₇	0.9706		0.38394	[7] SC
		0.97199(2)		0.38480(1)	^d
τ_{16} , Ce ₂ Rh ₁₅ Si ₇	<i>Pm3̄m</i> Ce ₂ Rh ₁₅ Si ₇	0.8818			[8] SC
		0.88323(2)			^d
τ_{17} , Ce ₃ Rh ₂ Si ₂	<i>Pbcm</i> La ₃ Ni ₂ Ga ₂	0.56762	0.79718	1.3271	[11] SC
					^d
τ_{18} , Ce ₄ Rh ₄ Si ₃	<i>C2/c</i> Sm ₄ Pd ₄ Si ₃	0.56849(5)	0.79745(7)	1.3289(1)	^d
		2.0749(1)	0.57242(3)	0.78741(3)	^d
τ_{19} , Ce ₃ Rh ₃ Si ₂	<i>Pnma</i> Ce ₃ Rh ₃ Si ₂	0.7722	1.4822	0.5759	[9] SC
		0.77036	1.4848	0.57466	[50]
τ_{20} , CeRh _{1.88} Si _{1.12}	<i>Cmcm</i> inv.-CeNiSi ₂	0.77176(6)	1.4874(1)	0.57599(5)	^d
		0.40591(3)	1.7673(1)	0.40736(3)	^d
τ_{21} , CeRh _{2-x} Si _{1+x}	<i>I4₁/amd</i> CeIr ₂ Si [64]	0.40413	1.7730	0.40675	SC for CeRh ₂ Si [10]
		0.40521(2)		3.5556(2)	0.0 ≤ <i>x</i> ≤ 0.1 ^d
τ_{22} , Ce ₂ Rh _{3+x} Si _{1-x}	<i>R3̄m</i> Y ₂ Rh ₃ Ge	0.40560(4)		3.5472(4)	<i>x</i> _{min} = 0 ^d
					<i>x</i> _{max} = 0.1 ^d
τ_{23} , Ce _{33.3} Rh _{58.2-55.2} Si _{8.5-11.5}	unknown oP* UMn ₂ (It)-deriv.	0.55577(2)		1.17766(6)	0.08 ≤ <i>x</i> ≤ 0.17 ^d
		0.55364(2)		1.19249(5)	<i>x</i> _{min} = 0.08 ^d
		1.5260(2)	0.7539(1)	0.5523(2)	<i>x</i> _{max} = 0.17 ^d
		1.5284(1)	0.7536(1)	0.5512(1)	Ce _{33.3} Rh _{55.2} Si _{11.5} ^d
τ_{24} , Ce ₈ (Rh _{1-x} Si _x) ₂₄ Si	<i>Pm3̄m</i> Ce ₈ Pd ₂₄ Sb	1.5354(3)	0.7530(1)	0.5496(1)	Ce _{33.3} Rh _{55.7} Si ₁₁ ^d
		0.5135(1)	0.7529(1)	0.5485(1)	Ce _{33.3} Rh _{57.7} Si ₉ ^{d,b}
τ_{25} , Ce ₄ (Rh _{1-x} Si _x) ₁₂ Si	<i>Im3̄m</i> Ce ₄ Rh ₁₂ Si	0.82166(3)			0.07 ≤ <i>x</i> ≤ 0.10 ^d
		0.82601(1)			<i>x</i> _{min} = 0.07 ^d
τ_{26} , Ce ₁₀ Rh ₅₁ Si ₃₃ ^c > 800 °C	<i>P6₃/m</i> U ₁₀ Co ₅₁ Si ₃₃	0.82155(1)			<i>x</i> _{max} = 0.10 ^d
		0.82342(1)			<i>x</i> _{min} = 0
τ_{27} , CeRhSi ^c > 800 °C	<i>P2₁3</i> LaIrSi	2.9107(1)		0.38334(1)	<i>x</i> _{max} = 0.03 ^d
		0.6231(1)			^d

^a Lattice parameters are given for BaAl₄-subcell.

^b Lattice parameters are given for UMn₂(It)-subcell.

^c Phase detected in as-cast alloy but does not participate in phase equilibria at 800 °C.

^d This work.

diffusion in the system combining elements with rather different melting points.

2. Experimental techniques

More than 80 alloys, each with a weight of 0.5 g, were prepared by argon arc-melting from high-purity elements (> 99.9 mass%) on a water-cooled copper hearth. To ensure homogenization, all alloys were re-melted three times. Part of each sample was vacuum-sealed in quartz tubes and annealed at 800 °C for 15–30 days before being quenched by dropping the capsules into cold water. In order to ensure that equilibrium conditions were achieved at this temperature, selected alloys were powdered, cold compacted and sintered at 800 °C for 1 week.

X-ray powder diffraction (XPD) data from as-cast and annealed alloys were collected from a Guinier-Huber 670 image plate system (CuK α ₁; 8° < 2 θ < 100°) and/or a STOE STADI P transmission diffractometer, equipped with a linear PSD (monochromatic CuK α ₁-radiation; 10° < 2 θ < 90°). Precise lattice parameters were calibrated against Ge as internal standard (*a*_{Ge} = 0.5657906 nm) using program STOE-WinXpow [52].

Single crystals were mechanically isolated from crushed alloys. Inspection on an AXS-GADDS texture goniometer assured high crystal quality, unit cell dimensions and Laue symmetry of the specimens prior to X-ray intensity data collection on a four-circle Nonius Kappa diffractometer equipped with a CCD area detector and employing graphite monochromated MoK α radiation (λ = 0.071073 nm). Orientation matrix and unit cell parameters were derived using program DENZO [53]. No absorption corrections were necessary because of the rather regular crystal shape and small dimensions of the investigated specimens. The structures were solved by direct methods [54] and refined with the SHELXL-97 program [55]. Quantitative X-ray Rietveld refinements were performed with the FULLPROF program [56,57], employing internal tables for X-ray atomic form factors. Atom parameters were standardized with the aid of program STRUCTURE TIDY [58].

All as-cast and annealed samples were polished via standard procedures and have been examined by scanning electron microscopy (SEM). Phase compositions were determined via electron probe microanalyses (EPMA) on a Carl Zeiss LEO EVO 50XVP instrument with a Link EDX INCA Energy 450 system (Q-BSD detector).

3. Binary systems

The Ce–Si binary phase diagram is used from a recent investigation by Bulanova et al. [59] amended by data on the existence of $\text{Ce}_2\text{Si}_{3-x}$ ($x=0.32$) [60]. We accepted the latest carefully re-investigated version of the Ce–Rh binary by Palenzona et al. [61]. The Rh–Si phase diagram is taken from the assessment of Schlesinger [62]. Crystallographic data pertinent to the Ce–Rh–Si system are summarized in Table 1. X-ray powder diffraction intensities for unary and binary phases reported in literature agree well with those observed in ternary Ce–Rh–Si alloys.

4. Results and discussion

4.1. Crystal structures of ternary phases

In the present investigation 25 ternary compounds were found to participate in the phase equilibria of the Ce–Rh–Si system at 800 °C (Fig. 1), 13 of which were already known earlier and were confirmed to exist by XPD phase analysis and EPMA measurements: τ_1 — CeRhSi_3 (BaNiSn_3 -type) [2,3], τ_2 — Ce_2RhSi_3 (Er_2RhSi_3 -type) [5], τ_3 — CeRhSi_2 (CeNiSi_2 -type) [34], τ_4 — $\text{Ce}_{22}\text{Rh}_{22}\text{Si}_{56}$ ($\text{Ce}_2\text{Rh}_2\text{Si}_5$ in [51], structure unknown), τ_6 — $\text{Ce}_2\text{Rh}_3\text{Si}_5$ ($\text{U}_2\text{Co}_3\text{Si}_5$ -type) [45,46], τ_9 — CeRh_2Si_2 (ThCr_2Si_2 -type) [1], τ_{12} — CeRh_3Si_2 (ErRh_3Si_2 -type) [4,48], τ_{14} — $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19}$ ($\text{Sc}_6\text{Co}_{30}\text{Si}_{19}$ -type) [7], τ_{15} — $\text{Ce}_2\text{Rh}_{12}\text{Si}_7$ ($\text{Ho}_2\text{Rh}_{12}\text{As}_7$ -type) [7], τ_{16} — $\text{Ce}_2\text{Rh}_{15}\text{Si}_7$ (own type) [8], τ_{19} — $\text{Ce}_3\text{Rh}_3\text{Si}_2$ (own type) [9], τ_{20} — $\text{CeRh}_{1.88}\text{Si}_{1.12}$ (inverse CeNiSi_2 -type) [10], and τ_{17} — $\text{Ce}_3\text{Rh}_2\text{Si}_2$ ($\text{La}_3\text{Ni}_2\text{Ga}_2$ -type) [11].

4.1.1. New compounds with known structure types

Based on XPD Rietveld refinements the crystal structures of nine new compounds were assigned to known structure types (Table 1). The structural chemistry of these compounds follows the characteristics already outlined for the prototype structures and thus will not be discussed here: τ_7 — Ce_3RhSi_3 ($\text{Ba}_3\text{Al}_2\text{Ge}_2$ -

type), τ_8 — $\text{Ce}_2\text{Rh}_{3-x}\text{Si}_{3+x}$ ($\text{Ce}_2\text{Rh}_{1.35}\text{Ge}_{4.65}$ -type), τ_{10} — $\text{Ce}_3\text{Rh}_{4-x}\text{Si}_{4+x}$ ($\text{U}_3\text{Ni}_4\text{Si}_4$ -type), τ_{11} — CeRh_6Si_4 (LiCo_6P_4 -type), τ_{13} — $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19.3}$ ($\text{U}_6\text{Co}_{30}\text{Si}_{19}$ -type), τ_{18} — $\text{Ce}_4\text{Rh}_4\text{Si}_3$ ($\text{Sm}_4\text{Pd}_4\text{Si}_3$ -type), τ_{21} — $\text{CeRh}_{2-x}\text{Si}_{1+x}$ (CeIr_2Si -type), τ_{22} — $\text{Ce}_2\text{Rh}_{3+x}\text{Si}_{1-x}$ ($\text{Y}_2\text{Rh}_3\text{Ge}$ -type) and τ_{24} — $\text{Ce}_8(\text{Rh}_{1-x}\text{Si}_x)_{24}\text{Si}$ ($\text{Ce}_8\text{Pd}_{24}\text{Sb}$ -type). The lattice parameters for these compounds and their homogeneity regions are summarized in Table 1. Results of Rietveld refinements of some of the aforementioned compounds are listed in Tables 2–5 with residual values being generally below $R_F < 0.059$: τ_8 — $\text{Ce}_2\text{Rh}_{3-x}\text{Si}_{3+x}$ (at $x=0.27$), τ_{10} — $\text{Ce}_3\text{Rh}_{4-x}\text{Si}_{4+x}$ (at $x=0$) and τ_{11} — CeRh_6Si_4 (Table 2); τ_{13} — $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19.3}$ (Table 3); τ_{17} — $\text{Ce}_3\text{Rh}_2\text{Si}_2$, τ_{18} — $\text{Ce}_4\text{Rh}_4\text{Si}_3$ and τ_{22} — $\text{Ce}_2\text{Rh}_{3+x}\text{Si}_{1-x}$ (at $x=0.15$) (Table 4); τ_{24} — $\text{Ce}_8(\text{Rh}_{1-x}\text{Si}_x)_{24}\text{Si}$ (at $x=0.09$) (Table 5).

The ternary phases τ_{11} – τ_{15} are forming in a narrow composition range (see inset in Fig. 1). EPMA analysis of the as-cast alloy $\text{Ce}_8\text{Rh}_{55}\text{Si}_{37}$ reveals four compositions (Fig. 2c), three of which were attributed to known phases by X-ray measurements: high temperature βRhSi , $\text{Rh}_{20}\text{Si}_{13}$ and τ_{11} — CeRh_6Si_4 . The fourth phase X (Fig. 2c) has the composition $\text{Ce}_{10.5}\text{Rh}_{54.4}\text{Si}_{35.1}$. In spite of proximity to τ_{14} — $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19}$, the X-ray pattern of $\text{Ce}_8\text{Rh}_{55}\text{Si}_{37}$ alloy does not show any trace of τ_{14} . Instead, reflections of two new compounds were detected: τ_{13} — $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19.3}$ and τ_{26} — $\text{Ce}_{10}\text{Rh}_{51}\text{Si}_{33}$ with structure types of $\text{U}_6\text{Co}_{30}\text{Si}_{19}$ and $\text{U}_{10}\text{Co}_5\text{Si}_{33}$, respectively. Together with the structures of τ_{14} — $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19}$ ($\text{Sc}_6\text{Co}_{30}\text{Si}_{19}$ -type), τ_{15} — $\text{Ce}_2\text{Rh}_{12}\text{Si}_7$ ($\text{Ho}_2\text{Rh}_{12}\text{As}_7$ -type) and the newly found phase τ_{11} — CeRh_6Si_4 (LiCo_6Si_4 -type) all these structures are members of a hexagonal structure series with linked triangular columns following the general formula $\text{Ce}_{n(n+1)}\text{Rh}_{6(n^2+1)}\text{Si}_{4n^2+3}$ described in [7,58] (for comparison of the structures see Table 6). For τ_{13} we first considered the SmRh_5Ge_3 -type [63] as possible structure type, but although Rietveld refinement results in a reliable R_F it reveals extremely short interatomic distances $d_{\text{Ce1-Rh1}}=0.224$ nm, $d_{\text{Rh7-Rh1}}=0.141$ nm and was therefore rejected. Analysis of the structure $\text{U}_6\text{Co}_{30}\text{Si}_{19}$ reveals an empty channel at $x=y=0$ with diameter 0.18781 nm for $\text{U}_6\text{Co}_{30}\text{Si}_{19}$ and 0.23296 nm for an isotopic atom

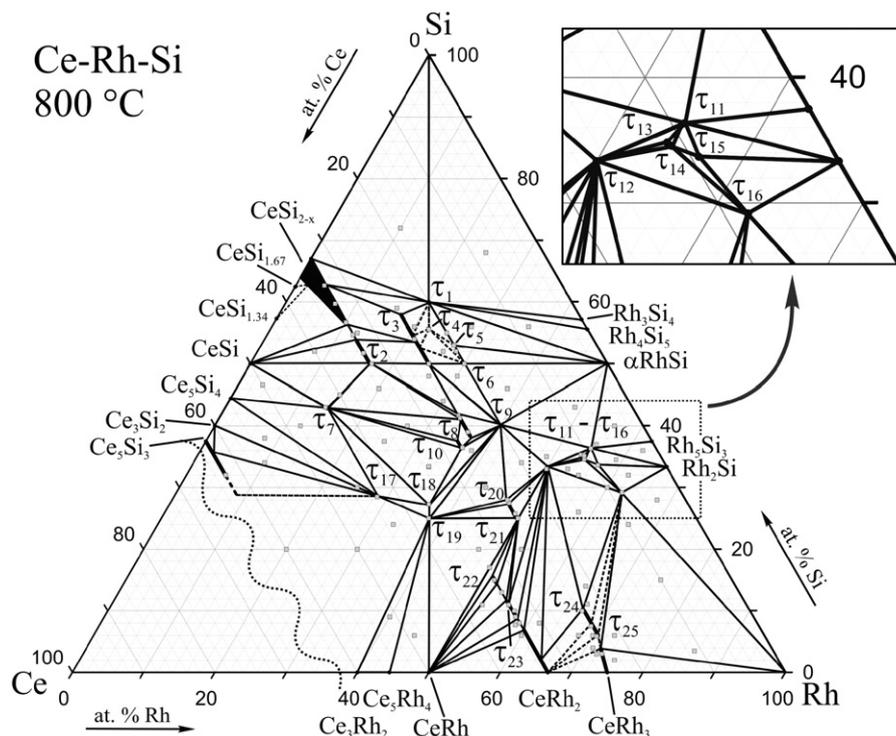


Fig. 1. Isothermal section of the ternary system Ce–Rh–Si at 800 °C. Gray squares indicates sample locations.

Table 2
Crystallographic data for phases τ_8 — $\text{Ce}_2\text{Rh}_3-x\text{Si}_{3+x}$ ($x=0.27$), τ_{10} — $\text{Ce}_3\text{Rh}_4-x\text{Si}_{4+x}$ ($x=0$) and τ_{11} — CeRh_6Si_4 (X-ray powder diffraction at room temperature, image plate, $\text{CuK}\alpha_1$ radiation, $20 \leq 2\theta \leq 100^\circ$).

Parameter/compound	τ_8 — $\text{Ce}_2\text{Rh}_3-x\text{Si}_{3+x}$	τ_{10} — $\text{Ce}_3\text{Rh}_4\text{Si}_4$	τ_{11} — CeRh_6Si_4
Composition, EPMA (at%)	$\text{Ce}_{25.1}\text{Rh}_{34.2}\text{Si}_{40.6}$	$\text{Ce}_{27.1}\text{Rh}_{36.2}\text{Si}_{36.6}$	$\text{Ce}_{9.0}\text{Rh}_{54.7}\text{Si}_{36.3}$
Composition from refinement (at%)	$\text{Ce}_{25.9}\text{Rh}_{34.1}\text{Si}_{40.9}$	$\text{Ce}_{27.2}\text{Rh}_{36.4}\text{Si}_{36.4}$	$\text{Ce}_{9.1}\text{Rh}_{54.5}\text{Si}_{36.4}$
Formula from refinement	$\text{Ce}_2\text{Rh}_{2.73}\text{Si}_{3.27}$	$\text{Ce}_3\text{Rh}_4\text{Si}_4$	CeRh_6Si_4
Space group	<i>Pmmn</i> (No. 59)	<i>Immm</i> (No. 71)	<i>P6mm</i> 2 (No. 187)
Pearson symbol	<i>oP16</i>	<i>oP22</i>	<i>hP11</i>
Prototype	$\text{Ce}_2\text{Rh}_{1.35}\text{Ge}_{4.65}$	$\text{U}_3\text{Ni}_4\text{Si}_4$	LiCo_6P_4
Lattice parameter (nm) (Ge standard)	$a=0.41103(1)$ $b=0.41047(1)$ $c=1.72595(2)$	$a=0.40676(1)$ $b=0.41405(1)$ $c=2.44689(5)$	$a=0.69851(1)$ $c=0.37823(1)$
Reflections measured	227	173	56
Number of variables	27	26	22
$R_F = \sum F_o - F_c / \sum F_o $	0.040	0.048	0.027
$R_I = \sum I_o - I_c / \sum I_o $	0.057	0.061	0.026
$R_{wp} = [\sum w_i y_{oi} - y_{ci} ^2 / \sum w_i y_{oi} ^2]^{1/2}$	0.046	0.040	0.033
$R_p = \sum y_{oi} - y_{ci} / \sum y_{oi} $	0.033	0.027	0.025
$R_e = [(N - P + C) / \sum w_i y_{oi}^2]^{1/2}$	0.020	0.017	0.015
$\chi^2 = (R_{wp} / R_e)^2$	5.4	5.5	5.1
<i>Atom parameters</i>			
Atom site 1	2 Ce1 in 2b (1/4,3/4,z); $z=0.8469(2)$	2 Ce1 in 2a (0,0,0)	1 Ce1 in 1a (0,0,0)
occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	1.64(4)	1.59(7)	0.92(2)
Atom site 2	2 Ce2 in 2a (1/4,1/4,z); $z=0.6458(3)$	4 Ce2 in 4j (1/2,0,z); $z=0.3543(1)$	3 Rh1 in 3k (x,–x, 1/2); $x=0.1998(1)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	1.64(4)	0.75(4)	0.57(3)
Atom site 3	2 Rh1 in 2b (1/4,3/4,z); $z=0.5022(4)$	4 Rh1 in 4j (1/2,0,z); $z=0.0997(1)$	3 Rh2 in 3j (x,–x,0); $x=0.5330(1)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	0.86(4)	1.19(6)	1.01(1)
Atom site 4	2 Rh2 in 2a (1/4,1/4,z); $z=0.0010(4)$	4 Rh2 in 4i (0,0,z); $z=0.2499(1)$	3 Si1 in 3k (x,–x, 1/2); $x=0.7984(3)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	0.86(4)	0.81(5)	1.49(1)
Atom site 5	2 Si1 in 2b (1/4,3/4,z); $z=0.0879(6)$	4 Si1 in 4j (1/2,0,z); $z=0.1966(2)$	1 Si1 in 1c (1/3,2/3,0)
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	0.86(4)	0.5(1)	1.49(1)
Atom site 6	2 Si2 in 2a (1/4,1/4,z); $z=0.4292(6)$	4 Si2 in 4i (0,0,z); $z=0.4589(3)$	
Occ.	1.00(–)	1.00(–)	
B_{iso} (10^2 nm^2)	0.86(4)	1.6(1)	
Atom site 7	2 M1 in 2b (1/4,3/4,z); $z=0.2062(4)$		
Occ.	Si 0.65(2)+Rh 0.35		
B_{iso} (10^2 nm^2)	0.86(4)		
Atom site 8	2 M2 in 2a (1/4,1/4,z); $z=0.2798(5)$		
Occ.	Si 0.62(2)+Rh 0.38		
B_{iso} (10^2 nm^2)	0.86(4)		

arrangement ‘ $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19}$ ’ (see Fig. 3a). In case of τ_{13} the channel is broad enough to provide space for an additional atom. A difference Fourier calculation for ‘ $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19}$ ’ in a partial y,z -section at $x=0$ (Fig. 3b) revealed an electron density $\sim 75 \text{ e}/\text{\AA}^3$ at the Wyckoff position 2b (0,0,0) and (0,0,1/2). Rietveld refinement for τ_{13} with an additional Si atom in the (0,0,0) position yielded an occupancy of 0.3(1) and $R_F=0.042$ (Table 3). The corresponding composition $\text{Ce}_6\text{Rh}_{30}\text{Si}_{19.3}$ is in good agreement with the EPMA value (Table 3).

The ternary compound τ_{20} , reported as stoichiometric CeRh_2Si [10] with the fully ordered and inverse orthorhombic CeNiSi_2 -type structure, unambiguously was detected at 800°C at a slightly non-stoichiometric composition $\text{CeRh}_{1.88}\text{Si}_{1.12}$ (Fig. 2e). However, at the stoichiometric composition 1:2:1 we detected a new

tetragonal structure (τ_{21}) isotypic to CeIr_2Si [64], which coexists in equilibrium with τ_{12} — CeRh_3Si_2 and τ_{23} — $\text{Ce}(\text{Rh}_{1-x}\text{Si}_x)_2$ (Fig. 2f) and has a small homogeneity range $\text{CeRh}_{2-x}\text{Si}_{1+x}$, $0.0 \leq x \leq 0.1$. A two-phase region between τ_{20} and τ_{21} was neither detected by EPMA nor from XPD analysis.

4.1.2. Single crystal structure determination of τ_3 — $\text{CeRh}_{1-x}\text{Si}_{2+x}$ and τ_6 — $\text{Ce}_2\text{Rh}_3\text{Si}_5$

Though the crystal structures of compounds CeRhSi_2 and $\text{Ce}_2\text{Rh}_3\text{Si}_5$ are known, neither single crystal nor powder data are available in literature.

A single crystal, broken from as-cast alloy with nominal composition $\text{Ce}_{20}\text{Rh}_{27}\text{Si}_{53}$ revealed orthorhombic symmetry with space group *Ibam* and lattice parameters: $a=0.99118(2)$,

Table 3Crystallographic data for τ_{13} -Ce₆Rh₃₀Si_{19.3} (X-ray powder diffraction at room temperature, image plate, CuK α_1 radiation, $20 \leq 2\theta \leq 100^\circ$).

Parameter/compound	τ_{13} -Ce ₆ Rh ₃₀ Si _{19.3}		
Composition, EPMA (at%)	Ce _{10.7} Rh _{54.0} Si _{35.3}		
Composition from refinement (at%)	Ce _{10.8} Rh _{54.2} Si _{34.9}		
Formula from refinement	Ce ₆ Rh ₃₀ Si _{19.3}		
Space group	P6 ₃ /m (no. 176)		
Pearson symbol	hP110		
Prototype	U ₆ Co ₃₀ Si ₁₉		
Lattice parameter (nm)	$a=2.2300(1)$		
(Ge standard)	$c=0.38398(1)$		
Reflections measured	728		
Number of variables	68		
$R_F = \sum F_o - F_c / \sum F_o $	0.042		
$R_I = \sum I_o - I_c / \sum I_o$	0.057		
$R_{wp} = [\sum w_i y_{oi} - y_{ci} ^2 / \sum w_i y_{oi} ^2]^{1/2}$	0.068		
$R_p = \sum y_{oi} - y_{ci} / \sum y_{oi} $	0.053		
$R_e = [(N - P + C) / \sum w_i y_{oi}^2]^{1/2}$	0.032		
$\chi^2 = (R_{wp}/R_e)^2$	4.6		
<i>Atom parameters</i>			
Atom sites	1 /8/ 15 6 Ce1 in 6h (x,y,1/4); $x=0.5238(5)$ $y=0.1351(5)$	6 Rh6 in 6h (x,y,1/4); $x=0.2027(6)$ $y=0.0536(6)$	6 Si3 in 6h (x,y,1/4); $x=0.124(2)$ $y=0.388(2)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10 ² nm ²)	0.42(2)	0.58(4) ^a	0.58(4) ^a
Atom sites	2 /9/ 16 6 Ce2 in 6h (x,y,1/4); $x=0.2640(4)$ $y=0.2297(5)$	6 Rh7 in 6h (x,y,1/4); $x=0.0454(6)$ $y=0.4914(5)$	6 Si4 in 6h (x,y,1/4); $x=0.761(2)$ $y=0.686(2)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10 ² nm ²)	1.44(2)	0.58(4) ^a	0.58(4) ^a
Atom sites	3 /10/ 17 6 Rh1 in 6h (x,y,1/4); $x=0.5486(5)$ $y=0.3055(6)$	6 Rh8 in 6h (x,y,1/4); $x=0.0064(5)$ $y=0.3579(6)$	6 Si5 in 6h (x,y,1/4); $x=0.069(2)$ $y=0.184(2)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10 ² nm ²)	0.58(4) ^a	0.58(4) ^a	0.58(4) ^a
Atom sites	4 /11/ 18 6 Rh2 in 6h (x,y,1/4); $x=0.8008(5)$ $y=0.2051(5)$	6 Rh9 in 6h (x,y,1/4); $x=0.3559(6)$ $y=0.6095(6)$	6 Si6 in 6h (x,y,1/4); $x=0.006(2)$ $y=0.579(2)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10 ² nm ²)	0.58(4) ^a	0.58(4) ^a	0.58(4) ^a
Atom sites	5 /12/ 19 6 Rh3 in 6h (x,y,1/4); $x=0.6573(6)$ $y=0.1040(5)$	6 Rh10 in 6h (x,y,1/4); $x=0.1048(5)$ $y=0.0960(5)$	2 Si7 in 2d (2/3,1/3,1/4)
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10 ² nm ²)	0.58(4) ^a	0.58(4) ^a	0.58(4) ^a
Atom sites	6 /13/ 20 6 Rh4 in 6h (x,y,1/4); $x=0.2891(7)$ $y=0.3917(6)$	6 Si1 in 6h (x,y,1/4); $x=0.471(2)$ $y=0.632(2)$	2 Si8 in 2b (0,0,0)
Occ.	1.00(–)	1.00(–)	0.3 (1)
B_{iso} (10 ² nm ²)	0.58(4) ^a	0.58(4) ^a	0.58(4) ^a
Atom sites	7/14 6 Rh5 in 6h (x,y,1/4); $x=0.1611(6)$ $y=0.3004(5)$	6 Si2 in 6h (x,y,1/4); $x=0.239(2)$ $y=0.470(2)$	
Occ.	1.00(–)	1.00(–)	
B_{iso} (10 ² nm ²)	0.58(4) ^a	0.58(4) ^a	

^a Constrained parameters.

$b=1.17558(3)$, $c=0.58313(1)$ nm. Direct methods yielded a completely ordered atom arrangement isotopic with the structure type of U₂Co₃Si₅. Results of the refinement for Ce₂Rh₃Si₅, which converged to $R_F=0.026$ are summarized in Table 7.

The X-ray diffraction pattern of a single crystal from the as-cast sample Ce_{2.5}Rh₁₆Si_{5.9} was indexed with orthorhombic symmetry: space group *Cmcm* and lattice parameters $a=0.42604(2)$, $b=1.67737(6)$, $c=0.41787(1)$ nm. The structure type CeNiSi₂ for τ_3 was confirmed by X-ray single crystal refinement. Atom parameters and results of the refinement for CeRh_{0.76}Si_{2.24}, which

converged to $R_F=0.017$ with residual electron densities smaller than $\pm 3.2 \text{ e}^-/\text{\AA}^3$, are summarized in Table 6.

4.2. Phase relations and isothermal section of the Ce–Rh–Si system at 800 °C

Table 8 summarizes compositions, structure type and lattice parameters for all the phases involved in three-phase equilibria at 800 °C. Besides the ternary compounds mentioned in Sections

Table 4
Crystallographic data for phases τ_{17} — $\text{Ce}_3\text{Rh}_2\text{Si}_2$, τ_{18} — $\text{Ce}_4\text{Rh}_4\text{Si}_3$ and τ_{22} — $\text{Ce}_2\text{Rh}_{3+x}\text{Si}_{1-x}$ ($x=0.15$) (X-ray powder diffraction at room temperature, image plate, $\text{CuK}\alpha_1$ radiation, $20 \leq 2\theta \leq 100^\circ$).

Parameter/compound	τ_{17} — $\text{Ce}_3\text{Rh}_2\text{Si}_2$	τ_{18} — $\text{Ce}_4\text{Rh}_4\text{Si}_3$	τ_{22} — $\text{Ce}_2\text{Rh}_{3+x}\text{Si}_{1-x}$
Composition, EPMA (at%)	$\text{Ce}_{42.6}\text{Rh}_{28.5}\text{Si}_{29.0}$	$\text{Ce}_{36.4}\text{Rh}_{36.6}\text{Si}_{27.0}$	$\text{Ce}_{33.2}\text{Rh}_{52.0}\text{Si}_{14.8}$
Composition from refinement (at%)	$\text{Ce}_{42.8}\text{Rh}_{28.6}\text{Si}_{28.6}$	$\text{Ce}_{36.4}\text{Rh}_{36.4}\text{Si}_{27.2}$	$\text{Ce}_{33.3}\text{Rh}_{52.4}\text{Si}_{14.3}$
Formula from refinement	$\text{Ce}_3\text{Rh}_2\text{Si}_2$	$\text{Ce}_4\text{Rh}_4\text{Si}_3$	$\text{Ce}_2\text{Rh}_{3.15}\text{Si}_{0.85}$
Space group	<i>Pbcm</i> (No. 57)	<i>C2/c</i> (No. 15)	<i>R3m</i> (No. 166)
Pearson symbol	<i>oP28</i>	<i>mC44</i>	<i>hR18</i>
Prototype	$\text{La}_3\text{Ni}_2\text{Ga}_2$	$\text{Sm}_4\text{Pd}_4\text{Si}_3$	$\text{Y}_2\text{Rh}_3\text{Ge}$
Lattice parameter (nm)	$a=0.56910(1)$	$a=2.07512(4)$	$a=0.55495(1)$
(Ge standard)	$b=0.79829(1)$	$b=0.57208(1)$	
	$c=1.32895(2)$	$c=0.78776(1)$	$c=1.18616(1)$
		$\beta=110.11(1)^\circ$	
Reflections measured	370	553	59
Number of variables	31	38	21
$R_F = \sum F_o - F_c / \sum F_o $	0.051	0.026	0.047
$R_I = \sum I_o - I_c / \sum I_o$	0.062	0.031	0.051
$R_{wp} = [\sum w_i y_{oi} - y_{ci} ^2 / \sum w_i y_{oi} ^2]^{1/2}$	0.034	0.013	0.036
$R_p = \sum y_{oi} - y_{ci} / \sum y_{oi} $	0.025	0.010	0.027
$R_e = [(N - P + C) / \sum w_i y_{oi}^2]^{1/2}$	0.018	0.009	0.015
$\chi^2 = (R_{wp} / R_e)^2$	3.6	2.3	5.9
<i>Atom parameters</i>			
Atom site 1	8 Ce1 in 8e (<i>x,y,z</i>); $x=0.1403(2)$ $y=0.3913(1)$ $z=0.1002(1)$	8 Ce1 in 8f (<i>x,y,z</i>); $x=0.0656(1)$ $y=0.1350(2)$ $z=0.6701(1)$	6 Ce1 in 6c (0,0,z); $z=0.3758(1)$
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	1.53(3)	1.13(3)	0.78(2)
Atom site 2	4 Ce2 in 4d (<i>x,y,1/4</i>); $x=0.6418(3)$ $y=0.2449(2)$	8 Ce2 in 8f (<i>x,y,z</i>); $x=0.3296(1)$ $y=0.8828(2)$ $z=0.0980(1)$	9 Rh1 in 9d (1/2,0,1/2)
Occ.	1.00(–)	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	0.55(4)	0.41(3)	1.03(2)
Atom site 3	8 Rh1 in 8e (<i>x,y,z</i>); $x=0.3773(4)$ $y=0.0347(1)$ $z=0.0921(1)$	8 Rh1 in 8f (<i>x,y,z</i>); $x=0.0606(1)$ $y=0.3635(3)$ $z=0.0189(2)$	3 M in 3a (0,0,0)
Occ.	1.00(–)	1.00(–)	0.86(1) Si + 0.14 Rh
B_{iso} (10^2 nm^2)	2.07(4)	1.51(4)	0.5(1)
Atom site 4	4 Si1 in 4d (<i>x,y,1/4</i>); $x=0.140(1)$ $y=0.1033(7)$	8 Rh2 in 8f (<i>x,y,z</i>); $x=0.2212(1)$ $y=0.1474(3)$ $z=0.8203(2)$	
Occ.	1.00(–)	1.00(–)	
B_{iso} (10^2 nm^2)	1.3(2)	1.56(4)	
Atom site 5	4 Si1 in 4c (<i>x,1/4,0</i>); $x=0.640(1)$	4 Si1 in 4e (0,y, 1/4); $y=0.359(1)$	
Occ.	1.00(–)	1.00(–)	
B_{iso} (10^2 nm^2)	2.9(2)	0.5(1)	
Atom site 6		8 Si2 in 8f (<i>x,y,z</i>); $x=0.1763(2)$ $y=0.8716(9)$ $z=0.5743(7)$	
Occ.		1.00(–)	
B_{iso} (10^2 nm^2)		0.37(9)	

4.1.1 and 4.1.2, we detected a ternary phase (τ_{23}) with unknown structure and composition range $\text{Ce}_{33.3}\text{Rh}_{58.2-55.2}\text{Si}_{8.5-11.5}$. The major reflections of the XPD pattern of τ_{23} may be indexed on the basis of an orthorhombic structure with $a_0 \approx 0.51 \text{ nm}$, $b_0 \approx 0.55 \text{ nm}$ and $c_0 \approx 0.75 \text{ nm}$, which suggests isotypism with the orthorhombic low-temperature modification of UMn_2 [65]. However, full profile analysis for this structure model did not result in a satisfactory description of the observed intensities. As the X-ray diffraction pattern reveals additional reflections, several samples with 33 at% of Ce were prepared. X-ray reflections of the phase in ternary samples with 10, 11 and 13 at% of Si were fully indexed on a supercell $a=3a_0$, $b=b_0$, $c=c_0$, unlike the super-

structure reflections of the phase in samples with 8 and 9 at% of Si. As the correct crystal structure is unavailable we are presently unable to decide whether the composition range of τ_{23} contains one or two phases.

Consistent with an incongruent formation of τ_{18} — $\text{Ce}_4\text{Rh}_4\text{Si}_3$ ($\text{Sm}_4\text{Pd}_4\text{Si}_3$ -type, see Table 2) from the melt, an as-cast sample $\text{Ce}_{36.5}\text{Rh}_{36.5}\text{Si}_{27}$ (composition close to $\text{Ce}_4\text{Rh}_4\text{Si}_3$) shows primary grains of τ_9 surrounded by secondary crystals of τ_8 whilst τ_{18} and binary Ce_3Rh_4 crystallize from the last portion of the liquid (see Fig. 2g). It is interesting to note, that ternary τ_{19} and τ_{10} , which both lie close to the described crystallization line, are not observed in the as-cast alloy, and τ_{10} appears only after anneal of

Table 5Crystallographic data for τ_{24} — $\text{Ce}_8(\text{Rh}_{1-x}\text{Si}_x)_{24}\text{Si}$ and τ_{25} — $\text{Ce}_4(\text{Rh}_{1-x}\text{Si}_x)_{12}\text{Si}$ (X-ray powder diffraction at room temperature, image plate, $\text{CuK}\alpha_1$ radiation, $20 \leq 2\theta \leq 100^\circ$).

Parameter/compound	τ_{24} — $\text{Ce}_8(\text{Rh}_{1-x}\text{Si}_x)_{24}\text{Si}$	τ_{25} — $\text{Ce}_4(\text{Rh}_{1-x}\text{Si}_x)_{12}\text{Si}$
Composition, EPMA (at%)	$\text{Ce}_{23.5}\text{Rh}_{66.5}\text{Si}_{10}$	$\text{Ce}_{23.5}\text{Rh}_{70.5}\text{Si}_6$
Composition from refinement (at%)	$\text{Ce}_{24.2}\text{Rh}_{66.4}\text{Si}_{9.6}$	$\text{Ce}_{23.5}\text{Rh}_{70.6}\text{Si}_{5.9}$
Formula from refinement	$\text{Ce}_8\text{Rh}_{21.9}\text{Si}_{3.1}$	$\text{Ce}_4\text{Rh}_{12}\text{Si}$
Space group	$Pm\bar{3}m$ (No 221)	$Im\bar{3}m$ (No 229)
Pearson symbol	$cP33$	$cI34$
Prototype	$\text{Ce}_8\text{Pd}_{24}\text{Sb}$	$\text{Ce}_4\text{Rh}_{12}\text{Si}$
Lattice parameter (nm) (Ge standard)	$a=0.82501(1)$	$a=0.82154(1)$
Reflections measured	100	43
Number of variables	19	18
$R_F = \sum F_o - F_c / \sum F_o $	0.059	0.035
$R_I = \sum I_o - I_c / \sum I_o$	0.048	0.029
$R_{WP} = [\sum w_i y_{oi} - y_{ci} ^2 / \sum w_i y_{oi} ^2]^{1/2}$	0.026	0.029
$R_p = \sum y_{oi} - y_{ci} / \sum y_{oi} $	0.019	0.021
$R_e = [(N - P + C) / \sum w_i y_{oi}^2]^{1/2}$	0.014	0.014
$\chi^2 = (R_{WP}/R_e)^2$	3.4	4.2
<i>Atom parameters</i>		
Atom site 1	8 Ce1 in 8g (x,x,x); $x=0.2483(1)$	8 Ce1 in 8c (1/4,1/4,1/4)
Occ.	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	0.78(2)	0.30(2)
Atom site 2	12 Rh1 in 12h (x,1/2,0); $x=0.2464(2)$	12 Rh1 in 12d (1/4,0,1/2)
Occ.	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	0.93(2)	0.91(3)
Atom site 3	6 M in 6e (x,0,0); $x=0.2940(2)$	12 Rh2 in 12e (x,0,0); $x=0.2729(1)$
Occ.	0.64(1) Rh+0.36 Si	1.00(–)
B_{iso} (10^2 nm^2)	1.71(6)	1.80(4)
Atom site 4	6 Rh2 in 6f (x, 1/2,1/2); $x=0.2345(2)$	2 Si1 in 2a (0,0,0);
Occ.	1.00(–)	1.00(–)
B_{iso} (10^2 nm^2)	2.18(3)	1.5(2)
Atom site 5	1 Si1 in 1a (0,0,0)	
Occ.	1.00(–)	
B_{iso} (10^2 nm^2)	1.7(2)	

this sample at 800 °C (Fig. 2h). Such a behavior indicates that τ_{10} does not form from the liquid but rather forms from a solid-state reaction.

The strong increase of the cell parameters for the ternary homogeneity region of CeRh_3 dissolving Si (Table 1) suggests an interstitial incorporation of Si atoms in the crystal structure. For alloy $\text{Ce}_{24}\text{Rh}_{73}\text{Si}_3$ Rietveld refinement of the CeRh_3 (Cu_3Au -type) structure with statistical occupation of Si in position (1/2,1/2,1/2) resulted in a slightly lower residual value $R_F=0.034$ ($\text{CeRh}_3\text{Si}_{1-x}$) than for Si/Rh substitution in the (0,1/2,1/2) position ($\text{Ce}(\text{Rh},\text{Si})_3$, $R_F=0.036$). With increasing Si content EPMA revealed a constant Ce concentration for the composition region $\text{Ce}_{23.5}\text{Rh}_{70.5-66.5}\text{Si}_{6-10}$ (23.5 ± 0.5 at%). The X-ray diffraction pattern for the alloy $\text{Ce}_{23.5}\text{Rh}_{66.5}\text{Si}_{10}$ was found to be very similar to $\text{CeRh}_3\text{Si}_{1-x}$, but additional reflections indicated a two-fold superstructure ($a \sim 2a_{\text{CeRh}_3}$). Whereas the XPD pattern of the sample with composition $\text{Ce}_{23.5}\text{Rh}_{69}\text{Si}_{7.5}$ shows the two-fold supercell reflections but with systematic extinctions $h+k+l=2n+1$ for a bcc structure, the XPD pattern of $\text{Ce}_{23.5}\text{Rh}_{66.5}\text{Si}_{10}$ reveals a cubic primitive two-fold superstructure. For a comparison of both patterns see Fig. 4. A structural model according to $\text{Ce}_8\text{Pd}_{24}\text{Sb}$ (SG $Pm\bar{3}m$) as proposed by Gordon and DiSalvo [66] was successfully employed to refine the primitive XPD pattern of $\text{Ce}_{23.5}\text{Rh}_{66.5}\text{Si}_{10}$ (results are presented in Table 2, $R_F=0.059$, $d_{\text{Si1}-\text{Rh2}}=0.2417$ nm). Consequently, Rietveld refinement for the bcc structure (at lower Si-contents) was based on the $\text{Ce}_8\text{Pd}_{24}\text{Sb}$ -type model ($\text{Ce}_8\text{Rh}_{24}\text{Si}_{1-x}$) with an additional Si-atom in (1/2,1/2,1/2) yielding

$R_F=0.035$ (Table 2, $d_{\text{Si1}-\text{Rh2}}=0.2242$ nm). For comparison of these structures with CeRh_3 see Fig. 5. In accordance to the existence of these two phases, the $\text{Ce}_{23.5}\text{Rh}_{70.5-66.5}\text{Si}_{6-10}$ region was split into two parts: τ_{24} — $\text{Ce}_8(\text{Rh}_{1-x}\text{Si}_x)_{24}\text{Si}$ ($0.07 \leq x \leq 0.10$) and τ_{25} — $\text{Ce}_4(\text{Rh}_{1-x}\text{Si}_x)_{12}\text{Si}$ ($0.0 \leq x \leq 0.03$), though a two-phase region was not detected by means of EPMA and XPD. It should be noted, that although there is a constant increase of cell parameters within the composition range $\text{Ce}_{23.5}\text{Rh}_{70.5-66.5}\text{Si}_{6-10}$, the structure of τ_{25} — $\text{Ce}_4(\text{Rh}_{1-x}\text{Si}_x)_{12}\text{Si}$ exhibits a Si atom in (1/2,1/2,1/2), whereas this site is vacant in Si-rich τ_{24} . However, Rietveld refinement for τ_{24} with a Si2 atom in (1/2,1/2,1/2) resulted in a worse $R_F=0.07$ and a very short distance $d_{\text{Si2}-\text{Rh2}}=0.2176$ nm. It should be furthermore mentioned, that in all XPD patterns of $\text{CeRh}_3\text{Si}_{1-x}$, τ_{24} and τ_{25} , two still unindexed reflections appear at $2\theta \approx 12.2^\circ$ and 14.1° (Fig. 4) with an intensity comparable to the (1 1 0) reflection and that the 2θ values of these reflections follow the changes in Si concentration. τ_{24} forms directly from the liquid and at 800 °C this phase is in equilibrium with τ_{12} and CeRh_2 (Fig. 2i and j).

A new phase τ_5 — $\text{Ce}_{20}\text{Rh}_{27}\text{Si}_{53}$ was detected between the compounds τ_1 — CeRhSi_3 and τ_6 — $\text{Ce}_2\text{Rh}_3\text{Si}_5$. The X-ray pattern of this phase looks very similar to τ_1 and reflections can be indexed on a tetragonal cell with parameters $a=0.41391(1)$ nm, $c=0.99650(2)$ nm, which lie between the parameters of τ_1 (BaNiSn_3 -type as an ordered variant of BaAl_4 -type; $a=0.42311$ (1) nm, $c=0.97816(3)$ nm) and τ_9 — CeRh_2Si_2 (ThCr_2Si_2 -type as another ordered variant of BaAl_4 -type; $a=0.40883(1)$ nm,

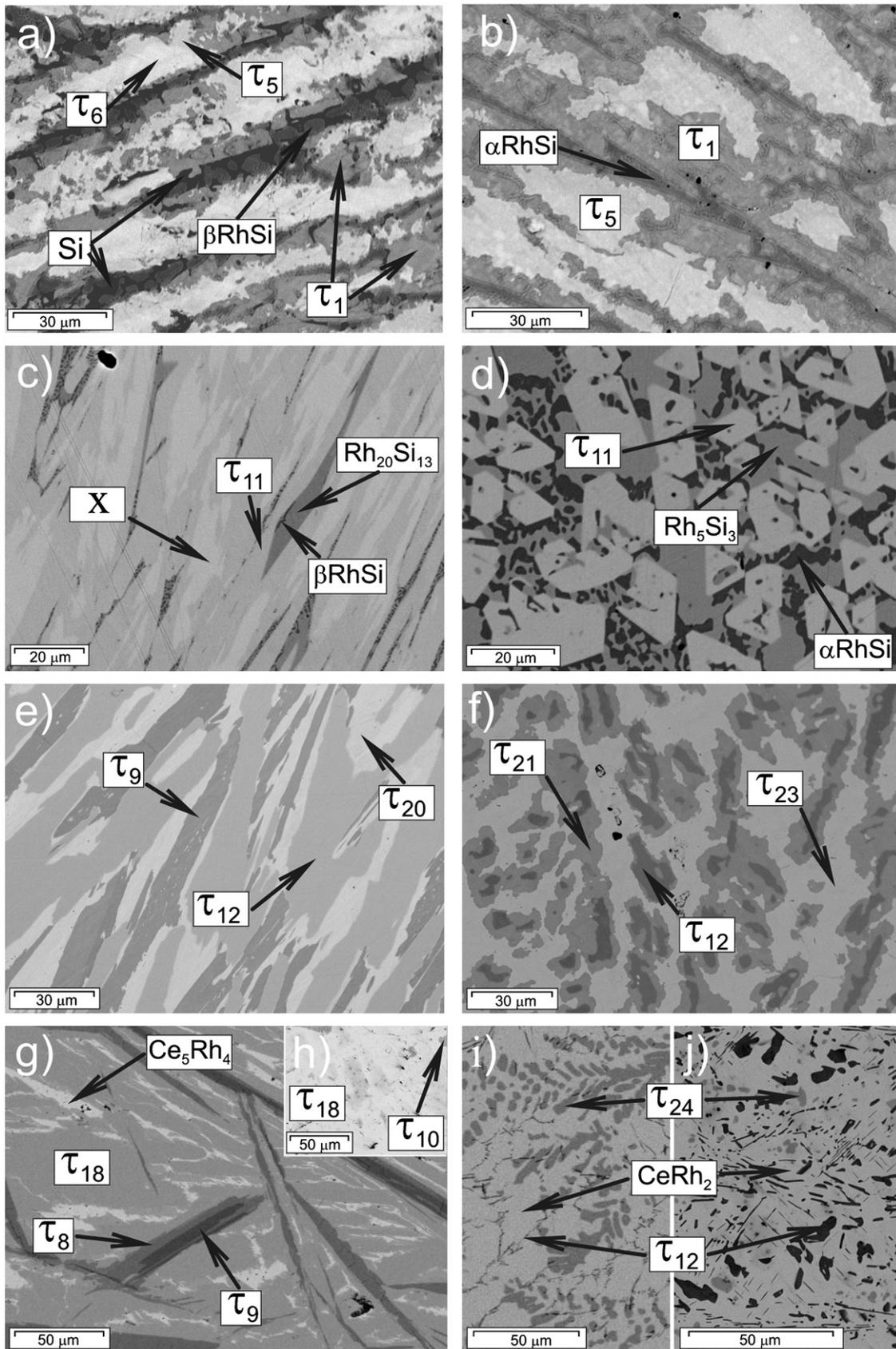


Fig. 2. Microstructure of selected Ce–Rh–Si alloys: $\text{Ce}_{16}\text{Rh}_{28}\text{Si}_{56}$ ((a) as-cast; (b) 800 °C), $\text{Ce}_8\text{Rh}_{55}\text{Si}_{37}$ ((c) as-cast, X— $\text{Ce}_{10.5}\text{Rh}_{54.4}\text{Si}_{35.1}$), $\text{Ce}_4\text{Rh}_{56}\text{Si}_{40}$ ((d) 800 °C), $\text{Ce}_{20}\text{Rh}_{46}\text{Si}_{34}$ ((e) 800 °C), $\text{Ce}_{27}\text{Rh}_{53}\text{Si}_{20}$ ((f) 800 °C), $\text{Ce}_{36.5}\text{Rh}_{36.5}\text{Si}_{27}$ ((g) as-cast; (h) 800 °C), $\text{Ce}_{30}\text{Rh}_{62}\text{Si}_8$ ((i) as-cast; (j) 800 °C).

Table 6
Hexagonal structures with $c \approx 0.38$ nm, which belong to the structure series with general formula $Ce_{n(n+1)}Rh_{6(n^2+1)}Si_{4n^2+3}$.

N	Compound	Structure type	Pearson symbol	Cell parameters (nm)		Crystal chemical formula
				a	c	
∞	τ_{11} —CeRh ₆ Si ₄	LiCo ₆ P ₄	hP11	0.69881(2)	0.37808(2)	CeRh ₆ Si ₄
1	τ_{15} —Ce ₂ Rh ₁₂ Si ₇	Ho ₂ Rh ₁₂ As ₇	hP24-3	0.97199(2)	0.38480(1)	Ce ₂ Rh ₁₂ Si ₇
2	τ_{14} —Ce ₆ Rh ₃₀ Si ₁₉	Sc ₆ Co ₃₀ Si ₁₉	hP62-7	1.57257(2)	0.38569(1)	Ce ₆ Rh ₃₀ Si ₁₉
3	τ_{13} —Ce ₆ Rh ₃₀ Si _{19.3}	U ₆ Co ₃₀ Si ₁₉	hP110	2.2300(1)	0.38398(1)	Ce ₁₂ Rh ₆₀ Si _{38.6}
4	τ_{25} —Ce ₁₀ Rh ₅₁ Si ₃₃	U ₁₀ Co ₅₁ Si ₃₃	hP188	2.9107(1)	0.38334(1)	Ce ₂₀ Rh ₁₀₂ Si ₆₆

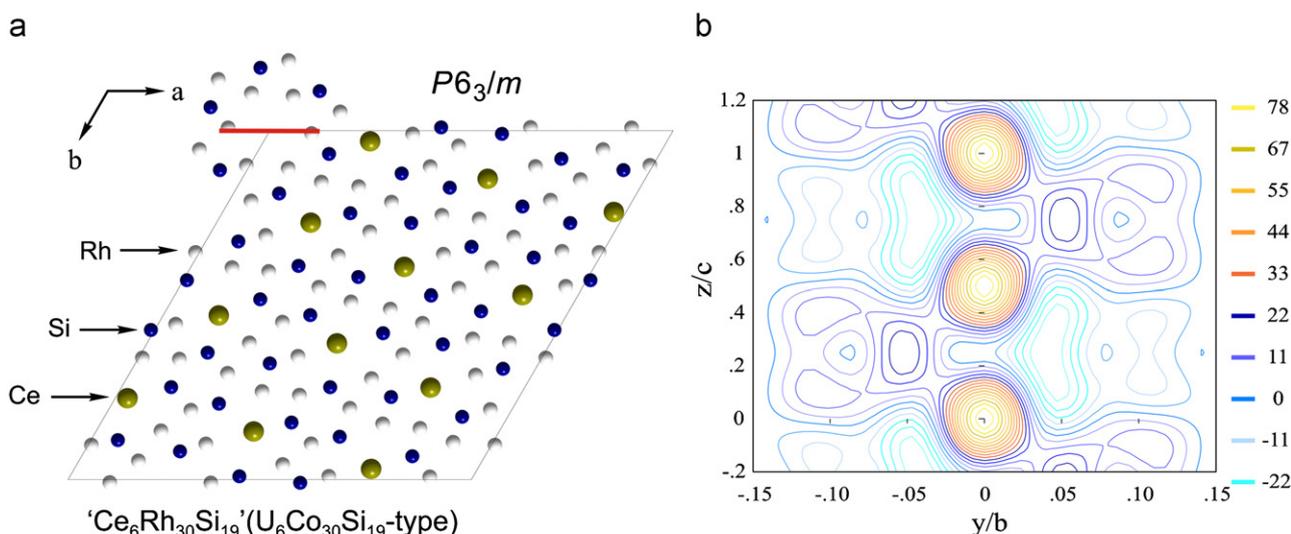


Fig. 3. Crystal structure of 'Ce₆Rh₃₀Si₁₉' (U₆Co₃₀Si₁₉-type) with empty channels at $x=y=0$ (a). Difference Fourier map for the structure along the partial section $x=0$; $-0.15 \leq y \leq 0.15$; $-0.2 \leq z \leq 1.2$ (marked by a thick solid line) (b).

$c = 1.01748(2)$ nm). Nevertheless the presence of additional reflections presently does not allow a structure designation for τ_5 . The phases τ_1 and τ_5 are, however, clearly distinguishable by BSD contrast (see Fig. 2a and b) and have a distinct grain boundary. The significant difference in the lattice parameters for these phases results in well separated X-ray patterns. In order to exclude the possibility that two different compositions are observed due to phase-segregation during crystallization from the liquid, the two-phase alloy Ce₂₀Rh₂₅Si₅₅ ($\tau_1 + \tau_5$) was subjected to heat treatment at 800 °C under various conditions: (a) anneal of the bulk as-cast alloy for 14 days, (b) sample was powdered to grain size below 20 μ m, cold compacted and sintered for 7 days, (c) powders were hot pressed at 800 °C for 2 h and annealed for 3 days. After all these treatments, the lattice parameters for τ_1 and τ_5 do not show any change that may indicate a single-phase region for these phases at 800 °C. Furthermore, the phases τ_1 , τ_5 and binary RhSi form a ternary equilibrium at 800 °C (Fig. 2b).

The annealing temperature of 800 °C was chosen in order to achieve enhanced diffusion in the system combining elements with rather different melting points. In general this temperature was enough for reaching equilibrium (see Fig. 2), but in some cases additional heat treatment was required. Thus more than three phases were detected by EPMA and XPD analysis for some samples after annealing at 800 °C for 1 month. Particularly it concerns equilibria involving τ_8 and τ_{10} phases. In these cases samples were powdered, cold pressed and annealed at 800 °C for 4 days. Equilibria for these regions were established from comparison of XPD patterns of the samples in different states.

Phase analysis of the samples prepared in the region of τ_4 (dashed lines) encountered similar problems, but due to unknown structure of the phase, solid evidence of phase equilibria cannot be presented. Moreover, EPMA of τ_4 shows a 2 at% spread of values of composition.

4.2.1. High- and low-temperature phases

The ternary compound τ_{26} —Ce₁₀Rh₅₁Si₃₃ was detected by means of XPD phase analysis in the as-cast alloy Ce_{10.7}Rh_{53.7}Si_{35.6} together with τ_{11} , τ_{12} and τ_{13} . However, a sample prepared at the stoichiometric composition Ce₁₀Rh₅₁Si₃₃ (=Ce_{10.6}Rh_{53.3}Si_{35.1}) neither contains the τ_{26} phase in as-cast nor in annealed states. We observed τ_{13} as a major phase with small amounts of τ_{11} and τ_{12} (and with Rh₂Si in the as-cast alloy). Because of the fact that during equilibration of the cast alloy Ce_{10.7}Rh_{53.7}Si_{35.6} at 800 °C the amount of τ_{26} —Ce₁₀Rh₅₁Si₃₃ significantly decreased and completely disappeared after annealing at 1050 °C, we consider τ_{26} to be a high temperature phase, which does not participate in the phase equilibria at 800 °C. It should be mentioned that due to the close compositions, EPMA is unable to distinguish between τ_{13} and τ_{26} phases.

Another high temperature phase—CeRhSi—was found in the as-cast sample Ce₃₃Rh₃₃Si₃₄ by means of EPMA analysis. The reflections of the unknown phase in the XPD pattern were indexed completely on the basis of a primitive cubic unit cell with cell parameter $a = 0.6231(1)$ nm and found to belong to the LaIrSi structure type. The phase completely disappeared after annealing at 800 °C as well as at 1050 °C.

Table 7X-ray single crystal data at RT for τ_3 —CeRh_{1-x}Si_{2+x} and τ_6 —Ce₂Rh₃Si₅ (MoK α radiation); structure data are standardized with program *Structure Tidy* [58].

Parameter/compound	τ_3 —CeRh _{1-x} Si _{2+x}	τ_6 —Ce ₂ Rh ₃ Si ₅
Nominal composition	CeRh _{0.76} Si _{2.24}	Ce ₂ Rh ₃ Si ₅
Crystal size (μm)	30 × 30 × 30	50 × 50 × 50
Space group	<i>Cmcm</i> (No. 63)	<i>Ibam</i> (No. 72)
Prototype	CeNiSi ₂	U ₂ Co ₃ Si ₅
Pearson symbol	<i>oS16</i>	<i>oI40</i>
Lattice parameters (nm)	<i>a</i> = 0.42604(2) <i>b</i> = 1.67737(6) <i>c</i> = 0.41787(1)	<i>a</i> = 0.99118(2) <i>b</i> = 1.17558(3) <i>c</i> = 0.58313(1)
Volume (nm ³)	0.29862(3)	0.67947(2)
μ_{abs} (mm ⁻¹)	21.04	21.03
2 θ range up to (deg)	72.34	72.49
Reflections in refinement	398 ≥ 4 $\sigma(F_o)$ of 424 −6 ≤ <i>h</i> ≤ 6 −25 ≤ <i>k</i> ≤ 27 −6 ≤ <i>l</i> ≤ 6	829 ≥ 4 $\sigma(F_o)$ of 863 −15 ≤ <i>h</i> ≤ 16 −19 ≤ <i>k</i> ≤ 19 −9 ≤ <i>l</i> ≤ 9
Index range		
Calculated density (g/cm ³)	6.66	7.13
Number of variables	19	37
$R_F = \sum F_o - F_c / \sum F_o $	0.017	0.026
R_{int}	0.011	0.012
wR^2	0.054	0.063
GOF	1.294	1.096
Extinction	0.0024(8)	0.0039(2)
<i>Atom parameters</i>		
Atom site 1	4 Ce1 in 4 <i>c</i> (0, <i>y</i> ,1/4); <i>y</i> = 0.39488(2)	8 Ce1 in 8 <i>j</i> (<i>x</i> , <i>y</i> ,0); <i>x</i> = 0.26612(3) <i>y</i> = 0.36895(2)
Occ.	1.00(−)	1.00(−)
U_{11}, U_{22}, U_{33} (in 10 ² nm ²)	0.0071(2), 0.0063(2), 0.0068(2)	0.0069(2), 0.0049(1), 0.0059(1)
U_{23}, U_{13}, U_{12} (in 10 ² nm ²)	0, 0, 0	0, 0, −0.0004(1)
Atom site 2	4 M1 in 4 <i>c</i> (0, <i>y</i> ,1/4); <i>y</i> = 0.18149(4)	8 Rh1 in 8 <i>j</i> (<i>x</i> , <i>y</i> ,0); <i>x</i> = 0.10789(4) <i>y</i> = 0.13811(3)
Occ.	0.762(4) Rh + 0.238 Si	1.00(−)
U_{11}, U_{22}, U_{33} (in 10 ² nm ²)	0.0072(3), 0.0075(3), 0.0058(3)	0.0061(2), 0.0053(2), 0.0086(2)
U_{23}, U_{13}, U_{12} (in 10 ² nm ²)	0, 0, 0	0, 0, 0.0001(1)
Atom site 3	4 Si1 in 4 <i>c</i> (0, <i>y</i> ,1/4); <i>y</i> = 0.0376(1)	4 Rh2 in 4 <i>b</i> (1/2,0,1/4)
Occ.	1.00(−)	1.00(−)
U_{11}, U_{22}, U_{33} (in 10 ² nm ²)	0.0094(9), 0.0121(9), 0.0079(9)	0.0059(2), 0.0057(2), 0.0064(2)
U_{23}, U_{13}, U_{12} (in 10 ² nm ²)	0, 0, 0	0, 0, 0
Atom site 4	4 Si2 in 4 <i>c</i> (0, <i>y</i> ,1/4); <i>y</i> = 0.7493(1)	8 Si1 in 8 <i>j</i> (<i>x</i> , <i>y</i> ,0); <i>x</i> = 0.3469(2) <i>y</i> = 0.1068(1)
Occ.	1.00(−)	1.00(−)
U_{11}, U_{22}, U_{33} (in 10 ² nm ²)	0.0095(8), 0.0066(7), 0.0069(8)	0.0165(8), 0.0082(6), 0.0075(6)
U_{23}, U_{13}, U_{12} (in 10 ² nm ²)	0, 0, 0	0, 0, 0.0030(5)
Atom site 5		8 Si2 in 8 <i>g</i> (0, <i>y</i> ,1/4); <i>y</i> = 0.2751(1)
Occ.		1.00(−)
U_{11}, U_{22}, U_{33} (in 10 ² nm ²)		0.0069(6), 0.0096(6), 0.0086(5)
U_{23}, U_{13}, U_{12} (in 10 ² nm ²)		0, 0.0012(4), 0
Atom site 5		4 Si3 in 4 <i>a</i> (0,0,1/4); <i>y</i> = 0.2751(1)
Occ.		1.00(−)
U_{11}, U_{22}, U_{33} (in 10 ² nm ²)		0.0058(8), 0.0061(8), 0.0091(8)
U_{23}, U_{13}, U_{12} (in 10 ² nm ²)		0, 0, 0
Residual density; e/Å ³ max; min	3.13; −1.12	8.02; −2.80
Principal mean square atomic displacements of U_{ij}	Ce1 0.0070 0.0068 0.0063 M1 0.0075 0.0072 0.0058 Si1 0.0121 0.0094 0.0079 Si2 0.0095 0.0069 0.0066	Ce1 0.0070 0.0059 0.0048 Rh1 0.0086 0.0061 0.0053 Rh2 0.0064 0.0059 0.0057 Si1 0.0175 0.0075 0.0073 Si2 0.0096 0.0092 0.0062 Si3 0.0091 0.0061 0.0058

Low-temperature binary Rh₃Si₂ was not formed during phase transformation in the Ce₄Rh₅₆Si₄₀ alloy (see Fig. 2d) confirming data of Palenzona et al. [61] on the metastable nature of this phase. Consequently the equilibrated alloy consists of α RhSi, Rh₅Si₃ and τ_{11} —CeRh₆Si₄.

5. Conclusions

Phase relations in the ternary system Ce–Rh–Si have been established for the isothermal section at 800 °C based on X-ray powder and single crystal diffraction, SEM and EMPA on about 80

Table 8
Data on alloys from three-phase regions in the Ce–Pd–Si system at 800 °C.

Three-phase field	Phase	EPMA (at%)			Lattice parameters (nm)		
		Ce	Rh	Si	a	b	c
(Si)+CeSi ₂ +τ ₁	(Si)	2.6	0.0	97.4	0.54226(3)		
	CeSi ₂	33.2	0.0	66.8	0.42007(2)		1.3943(1)
	τ ₁	19.4	20.4	60.2	0.42311(1)		0.97816(3)
(Si)+Rh ₃ Si ₄ +τ ₁	(Si)	0.0	0.0	100.0	0.54246(2)		
	Rh ₃ Si ₄	0.0	43.0	57.0	1.8800(1)	0.36048(2)	0.58089(3)
	τ ₁	19.4	20.7	59.9	0.42319(1)		0.97787(4)
CeSi ₂ +τ ₁ +τ ₃	CeSi ₂	33.4	3.6	63.0	0.41965(9)		1.3929(5)
	τ ₁	20.2	19.4	60.4	0.42327(5)		0.97837(2)
	τ ₃	24.9	16.6	58.5	0.42566(4)	1.6768(2)	0.41763(4)
τ ₁ +τ ₄ +τ ₅	τ ₁	20.0	20.2	59.8	0.42338(2)		0.98074(6)
	τ ₄	21.3	22.4	56.3	–	–	–
	τ ₅	19.8	27.3	52.9	0.41424(2)		0.99663(5)
αRhSi+τ ₁ +τ ₅	αRhSi	0.0	50.3	49.7	0.55466(5)	0.30707(2)	0.63658(5)
	τ ₁	19.3	20.8	59.9	0.42309(1)		0.97835(3)
	τ ₅	19.3	28.5	52.2	0.41346(1)		0.99404(5)
Rh ₄ Si ₅ +αRhSi+τ ₁	Rh ₄ Si ₅	0.0	44.3	55.7	1.23348(8)	0.35071(2)	0.59285(4)
	αRhSi	0.0	50.0	50.0	0.55467(7)	0.30773(3)	0.63583(7)
	τ ₁	19.9	20.2	59.9	0.42341(1)		0.97779(5)
CeSi+CeSi ₂ +τ ₂	CeSi	49.1	0.0	50.9	0.82678(7)	0.39704(3)	0.59768(4)
	CeSi ₂	33.6	9.4	57.0	0.41789(1)		1.41310(5)
	τ ₂	33.3	12.7	54.0	0.82068(1)		0.85030(2)
CeSi ₂ +τ ₃ +τ ₂	CeSi ₂	33.4	9.6	57.0	0.41781(1)		1.41377(2)
	τ ₃	25.3	21.5	53.2	0.42618(2)	1.67475(8)	0.41747(2)
	τ ₂	33.6	13.6	52.8	0.82162(1)		0.84804(2)
τ ₄ +τ ₅ +τ ₆	τ ₄	22.4	22.7	54.9	–	–	–
	τ ₅	19.9	26.9	53.2	0.41388(1)		0.99486(5)
	τ ₆	20.1	30.1	49.8	0.99003(3)	1.17412(4)	0.58187(2)
CeSi+Ce ₅ Si ₄ +τ ₇	CeSi	49.0	0.0	51.0	0.82832(3)	0.39756(1)	0.59601(2)
	Ce ₅ Si ₄	54.1	0.0	45.9	0.79482(7)		1.5143(2)
	τ ₇	42.1	14.5	43.4	0.41624(3)	0.42536(3)	1.82139(2)
CeSi+τ ₂ +τ ₇	CeSi	49.7	0.0	50.3	0.82876(9)	0.39694(4)	0.59685(7)
	τ ₂	33.7	16.7	49.6	0.82359(2)		0.84212(3)
	τ ₇	42.7	14.4	42.9	0.41528(1)	0.42548(1)	1.81906(5)
τ ₂ +τ ₇ +τ ₈	τ ₂	33.5	17.0	49.5	0.82283(2)		0.84149(3)
	τ ₇	42.4	14.5	43.1	0.41483(3)	0.42493(3)	1.8176(1)
	τ ₈	25.0	34.1	40.9	0.40968(3)	0.41094(3)	1.7221(1)
τ ₂ +τ ₃ +τ ₈	τ ₂	33.4	17.0	49.6	0.82403(2)		0.84280(3)
	τ ₃	24.7	25.2	50.1	0.42629(1)	1.67456(3)	0.41731(1)
	τ ₈	25.0	32.7	42.3	0.41243(2)	0.40980(1)	1.72551(7)
τ ₃ +τ ₈ +τ ₉	τ ₃	25.1	24.8	50.1	0.42562(3)	1.6716(1)	0.41673(3)
	τ ₈	25.0	32.7	42.3	0.41167(2)	0.40920(2)	1.72339(6)
	τ ₉	20.0	39.7	40.3	0.40836(1)		1.01565(3)
τ ₃ +τ ₆ +τ ₉	τ ₃	24.7	25.2	50.1	0.42613(1)	1.67520(5)	0.41705(1)
	τ ₆	20.0	30.2	49.8	0.98971(3)	1.17627(3)	0.58073(2)
	τ ₉	19.9	36.1	44.0	0.40922(1)		1.01504(4)
αRhSi+τ ₆ +τ ₉	αRhSi	0.0	51.2	48.8	0.55456(3)	0.30704(2)	0.63622(3)
	τ ₆	19.8	30.6	49.6	0.98862(4)	1.17548(4)	0.58034(2)
	τ ₉	19.7	39.8	40.5	0.40848(1)		1.01582(3)
αRhSi+τ ₉ +τ ₁₁	αRhSi	0.0	50.6	49.4	0.55496(2)	0.30721(1)	0.63700(2)
	τ ₉	19.2	40.5	40.3	0.40889(1)		1.01657(4)
	τ ₁₁	8.4	54.9	36.7	0.69833(2)		0.37833(1)
αRhSi+Rh ₅ Si ₃ +τ ₁₁	αRhSi	0.0	50.8	49.2	0.55517(3)	0.30707(1)	0.63736(3)
	Rh ₅ Si ₃	0.0	62.6	37.4	0.53181(1)	1.01257(2)	0.38966(1)
	τ ₁₁	9.1	54.3	36.6	0.69846(1)		0.37819(1)
Ce ₅ Si ₄ +Ce ₃ Si ₂ +τ ₁₇	Ce ₅ Si ₄	55.4	0.0	44.6	0.79385(3)		1.5072(1)
	Ce ₃ Si ₂	60.0	0.0	40.0	0.77908(4)		0.43690(3)
	τ ₁₇	42.6	29.0	28.4	0.56866(3)	0.79739(5)	1.3278(1)
Ce ₅ Si ₄ +τ ₇ +τ ₁₇	Ce ₅ Si ₄	55.0	0.0	45.0	0.79438(3)		1.5098(1)
	τ ₇	43.0	14.1	42.9	0.41567(2)	0.42595(2)	1.8175(1)
	τ ₁₇	42.3	28.6	29.1	0.56840(3)	0.79817(4)	1.32864(6)
τ ₇ +τ ₁₀ +τ ₁₈	τ ₇	42.4	14.6	43.0	0.41553(2)	0.42573(2)	1.81848(8)
	τ ₁₀	26.8	36.5	36.7	0.40683(2)	0.41365(2)	2.4462(1)
	τ ₁₈	35.8	36.9	27.3	2.0749(1)	0.57242(3)	0.78741(3)
τ ₇ +τ ₈ +τ ₁₀	τ ₇	42.6	14.5	42.9	0.41470(2)	β=110.09(1) [°] 0.42472(2)	1.83269(7)
	τ ₈	24.8	34.0	41.2	0.41140(1)	0.41038(1)	1.72547(5)
	τ ₁₀	27.4	35.3	37.3	0.40813(2)	0.41322(2)	2.4465(1)
τ ₈ +τ ₉ +τ ₁₀	τ ₈	25.2	36.7	38.1	0.41035(2)	0.41091(2)	1.72633(7)
	τ ₉	20.2	40.3	39.5	0.40883(1)		1.01748(2)
	τ ₁₀	27.1	36.2	36.7	0.40752(2)	0.41374(2)	2.4466(1)
τ ₉ +τ ₁₀ +τ ₁₈	τ ₉	20.0	40.6	39.4	0.40884(1)		1.01759(2)
	τ ₁₀	27.1	36.7	36.2	0.40591(3)	0.41438(3)	2.4464(2)
	τ ₁₈	36.2	36.4	27.4	2.0747(1)	0.57210(2)	0.78697(3)
τ ₉ +τ ₁₉ +τ ₂₀	τ ₉	19.8	40.9	39.3	0.40876(1)		1.01765(3)
	τ ₁₉	36.9	37.9	25.2	0.77171(3)	1.48725(6)	0.57739(2)
	τ ₂₀	24.7	48.9	26.4	0.40568(2)	1.76691(7)	0.40727(2)

Table 8 (continued)

Three-phase field	Phase	EPMA (at%)			Lattice parameters (nm)		
		Ce	Rh	Si	a	b	c
$\tau_9 + \tau_{12} + \tau_{20}$	τ_9	19.9	40.8	39.3	0.40871(1)		1.01731(2)
	τ_{12}	17.0	49.9	33.1	0.71252(1)	0.97220(2)	0.55932(1)
	τ_{20}	24.7	46.6	28.7	0.40618(1)	1.76413(5)	0.40794(1)
$\tau_9 + \tau_{11} + \tau_{12}$	τ_9	20.1	39.8	40.1	0.40892(1)		1.01691(4)
	τ_{11}	9.1	54.6	36.3	0.69899(5)		0.37922(4)
	τ_{12}	17.1	50.0	32.9	0.71273(2)	0.97240(3)	0.55935(2)
$\tau_{11} + \tau_{12} + \tau_{13}$	τ_{11}	9.1	54.6	36.3	0.69881(2)		0.37808(2)
	τ_{12}	16.4	50.4	33.2	0.71268(4)	0.97226(6)	0.55981(4)
	τ_{13}	10.9	54.4	34.7	2.23250(3)		0.38427(1)
$\tau_{12} + \tau_{14} + \tau_{16}$	τ_{12}	16.7	50.4	32.9	0.71289(1)	0.97255(2)	0.55960(1)
	τ_{14}	10.9	54.5	34.6	1.57257(2)		0.38569(1)
	τ_{16}	8.7	62.6	28.7	0.88329(2)		
$\text{Rh}_5\text{Si}_3 + \text{Rh}_2\text{Si} + \tau_{11}$	Rh_5Si_3	0.0	62.7	37.3	0.53179(1)	1.01235(2)	0.38958(1)
	Rh_2Si	0.0	66.5	33.5	0.54138(1)	0.39286(1)	0.73788(2)
	τ_{11}	9.0	54.8	36.2	0.69853(2)		0.37767(1)
$\text{Rh}_2\text{Si} + \tau_{11} + \tau_{15}$	Rh_2Si	0.0	66.6	33.4	0.54184(2)	0.39281(2)	0.73812(3)
	τ_{11}	8.7	55.1	36.2	0.69869(2)		0.37820(1)
	τ_{15}	9.2	57.2	33.6	0.97233(3)		0.38448(1)
$\text{Rh}_2\text{Si} + \tau_{15} + \tau_{16}$	Rh_2Si	0.0	67.2	32.8	0.54257(1)	0.39254(1)	0.73800(2)
	τ_{15}	9.3	57.3	33.4	0.97199(2)		0.38480(1)
	τ_{16}	8.4	62.5	29.1	0.88323(2)		
$\text{Ce}_5\text{Rh}_4 + \text{CeRh} + \tau_{19}$	Ce_5Rh_4	54.6	45.4	0.0	0.7486(1)	1.4788(2)	0.7657(1)
	CeRh	49.5	50.5	0.0	0.38528(4)	1.0975(1)	0.41537(4)
	τ_{19}	37.1	38.0	24.9	0.77153(3)	1.48800(7)	0.57637(3)
$\text{CeRh} + \tau_{19} + \tau_{21}$	CeRh	49.8	50.2	0.0	0.38786(7)	1.0888(2)	0.41517(7)
	τ_{19}	36.9	38.2	24.9	0.77176(6)	1.4874(1)	0.57599(5)
	τ_{21}	24.7	50.1	25.2	0.40564(2)		3.5533(3)
$\text{CeRh} + \tau_{21} + \tau_{22}$	CeRh	49.1	50.9	0.0	0.38439(7)	1.0946(2)	0.41739(7)
	τ_{21}	24.7	50.6	24.7	0.40498(2)		3.5535(3)
	τ_{22}	32.8	51.7	15.5	0.55577(2)		1.17766(6)
$\tau_{21} + \tau_{22} + \tau_{23}$	τ_{21}	–	–	–	–	–	–
	τ_{22}	33.5	52.7	13.8	0.55364(2)		1.19249(5)
	τ_{23}	33.4	55.1	11.5	1.5260(2)	0.7539(1)	0.5523(2)
$\text{CeRh} + \text{CeRh}_2 + \tau_{23}$	CeRh	–	–	–	–	–	–
	CeRh_2	33.0	60.3	6.7	0.75220(5)		–
	τ_{23}	33.4	58.2	8.4	1.5431(2)	0.7531(1)	0.5480(1)
$\text{CeRh}_2 + \tau_{24} + \tau_{12}$	CeRh_2	32.6	64.5	2.9	0.75256(2)		–
	τ_{24}	23.2	66.5	10.3	0.82524(1)		–
	τ_{12}	16.2	50.4	33.4	0.71264(7)	0.9707(2)	0.5591(1)
$\tau_{24} + \tau_{12} + \tau_{16}$	τ_{24}	22.5	67.3	10.2	0.82548(1)		–
	τ_{12}	16.8	50.2	33.0	0.71260(2)	0.97160(3)	0.55913(2)
	τ_{16}	8.3	62.8	28.9	0.88242(2)		–
$\text{Rh} + \text{CeRh}_3 + \tau_{16}$	(Rh)	0.0	100.0	0.0	0.38011(1)		–
	CeRh_3	24.3	71.1	4.6	0.40900(1)		–
	τ_{16}	8.2	63.0	28.8	0.88225(1)		–
$\text{Rh} + \text{Rh}_2\text{Si} + \tau_{16}$	(Rh)	0.0	98.0	2.0	0.38016(1)		–
	Rh_2Si	0.0	66.7	33.3	0.54109(1)	0.39294(1)	0.73855(1)
	τ_{16}	8.4	63.3	28.3	0.88266(1)		–

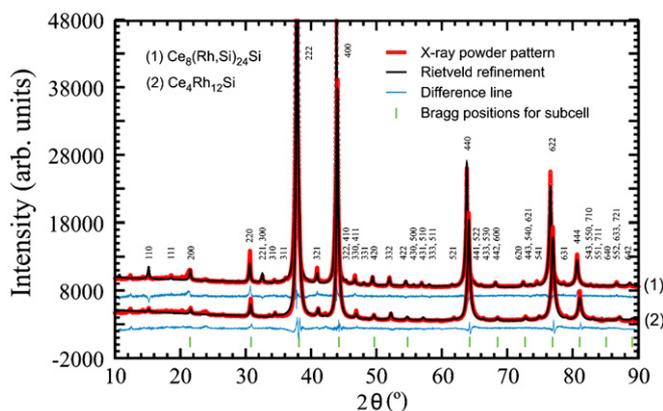


Fig. 4. XPD pattern for samples $\text{Ce}_{23.5}\text{Rh}_{66.5}\text{Si}_{10}$ (1) and $\text{Ce}_{23.5}\text{Rh}_{70.5}\text{Si}_6$ (2) with Rietveld refinement for primitive $\text{Ce}_8(\text{Rh,Si})_{24}\text{Si}$ -type and bcc $\text{Ce}_4\text{Rh}_{12}\text{Si}$ -type, respectively. Bragg positions are given for the CeRh_3 subcell.

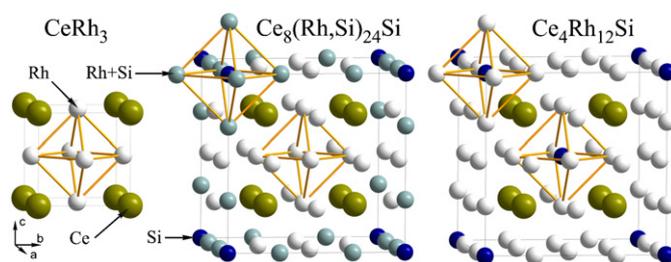


Fig. 5. Increasing Si content starting from binary CeRh_3 (Au_3Cu -type, left panel) yields a two-fold superstructure at low Si-contents with bcc symmetry ($\text{Ce}_4\text{Rh}_{12}\text{Si}$, right panel) and a primitive cubic structure at higher Si-concentrations ($\text{Ce}_8(\text{Rh,Si})_{24}\text{Si}$, middle panel) (for details see text).

alloys, which were prepared by various methods employing arc melting under argon or powder reaction sintering. Twenty-seven ternary compounds were detected in the system, and 25 of them

participate in the phase equilibria at 800 °C. Structure prototypes were assigned to nine new Ce–Rh–Si compounds. Phase equilibria are characterized by the absence of cerium solubility in the various rhodium silicides. However, mutual solubilities among cerium silicides and cerium–rhodium compounds are significant.

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Appendix A. Supplementary materials

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.jssc.2010.01.029.

References

- [1] R. Ballestracci, C. R. Acad. Sci. Ser. B 282 (13) (1976) 291.
- [2] P. Lejay, I. Higashi, B. Chevalier, J. Etourneau, P. Hagenmuller, Mater. Res. Bull. 19 (1) (1984) 115.
- [3] P. Haen, P. Lejay, B. Chevalier, B. Lloret, J. Etourneau, M. Sera, J. Less-Common Met. 110 (1–2) (1985) 321.
- [4] B. Chevalier, P. Lejay, J. Etourneau, P. Hagenmuller, Mater. Res. Bull. 18 (3) (1983) 315.
- [5] B. Chevalier, P. Lejay, J. Etourneau, P. Hagenmuller, Solid State Commun. 49 (8) (1984) 753.
- [6] B.I. Shapiev, Interaction of the elements and properties of the alloys in the ternary systems Ce–{Ru, Rh}–{Si, Ge}, Ph.D. Thesis, Moscow State University, 1993.
- [7] A.I. Tursina, A.V. Gribanov, Y.D. Seropegin, O.I. Bodak, J. Alloys Compd. 367 (1–2) (2004) 142.
- [8] A.I. Tursina, A.V. Gribanov, Y.D. Seropegin, Acta Crystallogr. E 57 (7) (2001) i55.
- [9] Y.M. Prots, J. Stepien-Damm, P.S. Salamakha, O.I. Bodak, J. Alloys Compd. 256 (1–2) (1997) 166.
- [10] A.I. Tursina, A.V. Gribanov, Y.D. Seropegin, A.A. Novitskii, O.I. Bodak, J. Alloys Compd. 367 (1–2) (2004) 146.
- [11] A.I. Tursina, A.V. Gribanov, Y.D. Seropegin, 2010, to be published.
- [12] K. Cenozual, B. Chabot, E. Parthe, Acta Crystallogr. C 44 (2) (1988) 221.
- [13] R.E. Gladyshevskii, K. Cenozual, E. Parthe, J. Alloys Compd. 189 (2) (1992) 221.
- [14] N. Kimura, Y. Muro, H. Aoki, J. Phys. Soc. Jpn. 76 (5) (2007).
- [15] N. Aso, H. Miyano, H. Yoshizawa, N. Kimura, T. Komatsubara, H. Aoki, J. Magn. Mater. 310 (2, Part 1) (2007) 602.
- [16] N. Kimura, K. Ito, H. Aoki, S. Uji, T. Terashima, Phys. Rev. Lett. 98 (19) (2007) 197001/1.
- [17] Y. Muro, M. Ishikawa, K. Hirota, Z. Hiroi, N. Takeda, N. Kimura, H. Aoki, J. Phys. Soc. Jpn. 76 (3) (2007) 033706.
- [18] T. Shimoda, Y. Okuda, Y. Takeda, Y. Ida, Y. Miyauchi, T. Kawai, T. Fujie, I. Sugitani, A. Thamizhavel, T.D. Matsuda, Y. Haga, T. Takeuchi, M. Nakashima, R. Settai, Y. Onuki, J. Magn. Mater. 310 (2) (2007) 308.
- [19] T. Terashima, Y. Takahide, T. Matsumoto, S. Uji, N. Kimura, H. Aoki, H. Harima, Phys. Rev. B 76 (5) (2007) 054506.
- [20] T. Terashima, M. Kimata, S. Uji, T. Sugawara, N. Kimura, H. Aoki, H. Harima, Phys. Rev. B 78 (20) (2008) 205107.
- [21] C. Godart, L.C. Gupta, M.F. Ravet Krill, J. Less-Common Met. 94 (1) (1983) 187.
- [22] B.H. Grier, J.M. Lawrence, V. Murgai, R.D. Parks, Phys. Rev. B 29 (1984) 2664.
- [23] S. Quezel, J. Rossat-Mignod, B. Chevalier, P. Lejay, J. Etourneau, Solid State Commun. 49 (7) (1984) 685.
- [24] R. Movshovich, T. Graf, D. Mandrus, J.D. Thompson, J.L. Smith, Z. Fisk, Phys. Rev. B 53 (13) (1996) 8241.
- [25] R. Settai, A. Misawa, S. Araki, M. Kosaki, K. Sugiyama, T. Takeuchi, K. Kindo, Y. Haga, E. Yamamoto, Y. Onuki, J. Phys. Soc. Jpn. 66 (8) (1997) 2260.
- [26] H. Abe, H. Kitazawa, H. Suzuki, G. Kido, T. Matsumoto, J. Magn. Mater. 177–181 (Part 1) (1998) 479.
- [27] H. Mori, N. Takeshita, N. Mori, Y. Uwatoko, Physica B 259–261 (1999) 58.
- [28] T. Muramatsu, S. Eda, T.C. Kobayashi, M.I. Erements, K. Amaya, S. Araki, R. Settai, Y. Onuki, Physica B 259–261 (1999) 61.
- [29] S. Kawarazaki, M. Sato, Y. Miyako, N. Chigusa, K. Watanabe, N. Metoki, Y. Koike, M. Nishi, Phys. Rev. B 61 (6) (2000) 4167.
- [30] S. Araki, R. Settai, T.C. Kobayashi, H. Harima, Y. Onuki, Phys. Rev. B 64 (22) (2001) 224417/1.
- [31] S. Araki, M. Nakashima, R. Settai, T.C. Kobayashi, Y. Onuki, J. Phys.: Condens. Matter 14 (21) (2002) L377.
- [32] M. Ohashi, G. Oomi, S. Koiwai, M. Hedо, Y. Uwatoko, Phys. Rev. B 68 (14) (2003) 144428.
- [33] A. Villaume, D. Aoki, Y. Haga, G. Knebel, R. Boursier, J. Flouquet, J. Phys.: Condens. Matter 20 (1) (2008) 015203.
- [34] B. Chevalier, P. Rogl, K. Hiebl, J. Etourneau, J. Solid State Chem. 107 (2) (1993) 327.
- [35] D.T. Adroja, B.D. Rainford, J. Magn. Mater. 119 (1–2) (1993) 54.
- [36] J.J. Lu, M.K. Lee, Y.M. Lu, L.Y. Jang, J. Magn. Mater. 311 (2) (2007) 614.
- [37] A. Szytula, J. Leciejewicz, K. Maletka, J. Magn. Mater. 118 (3) (1993) 302.
- [38] I. Das, E.V. Sampathkumaran, J. Magn. Mater. 137 (3) (1994) L239.
- [39] J. Leciejewicz, N. Stüsser, A. Szytula, A. Zygmunt, J. Magn. Mater. 147 (1–2) (1995) 45.
- [40] T. Nakano, K. Sengupta, S. Rayaprol, M. Hedо, Y. Uwatoko, E.V. Sampathkumaran, J. Phys.: Condens. Matter 19 (32) (2007) 326205.
- [41] S. Patil, K.K. Iyer, K. Maiti, E.V. Sampathkumaran, Phys. Rev. B 77 (9) (2008) 094443.
- [42] M. Szlowska, D. Kaczorowski, T. Plackowski, L.D. Gulay, Acta Phys. Pol. A 115 (1) (2009) 132.
- [43] Y. Muro, S. Takahashi, K. Sunahara, K. Motoya, M. Akatsu, N. Shirakawa, J. Magn. Mater. 310 (2007) e40.
- [44] D. Kaczorowski, in: X International Conference on Crystal Chemistry of Intermetallic Compounds, Lviv, Ukraine, 2007, Book of Abstracts, p. 7.
- [45] C. Godart, L.C. Gupta, C.V. Tomy, S. Patil, R. Nagarajan, E. Beaurepaire, R. Vijayaraghavan, J.V. Yakhmi, Mater. Res. Bull. 23 (1988) 1781.
- [46] S. Ramakrishnan, N.G. Patil, A.D. Chinchure, V.R. Marathe, Phys. Rev. B 64 (6) (2001) 064514.
- [47] T. Graf, M.F. Hundley, R. Modler, R. Movshovich, J.D. Thompson, D. Mandrus, R.A. Fisher, N.E. Phillips, Phys. Rev. B 57 (13) (1998) 7442.
- [48] D. Kaczorowski, T. Komatsubara, Physica B 403 (5–9) (2008) 1362.
- [49] A.P. Pikul, D. Kaczorowski, Acta Phys. Pol. A 115 (1) (2009) 235.
- [50] D. Kaczorowski, Y. Prots, U. Burkhardt, Y. Grin, Intermetallics 15 (3) (2007) 225.
- [51] D. Kaczorowski, in: 16th International Conference on Solid Compounds of Transition Elements, Dresden, Germany, 2008, Book of Abstracts, p. 312.
- [52] STOE WINXPOW (Version 1.06), Stoe & Cie GmbH, Darmstadt, Germany, 1999.
- [53] Nonius, Kappa CCD Program Package: COLLECT, DENZO, SCALEPACK, SORTAV, Nonius BV, Delft, The Netherlands, 1998.
- [54] G.M. Sheldrick, Acta Crystallogr. A 46 (1990) 467.
- [55] SHELXL-97. Program Crystal Structure Refinement, University of Gottingen, Germany, 1997.
- [56] J. Rodriguez-Carvajal, in: Satellite Meeting on Powder Diffraction of the XV Congress of the IUCr, Toulouse, France, 1990, Book of Abstracts, p. 127.
- [57] T. Roisnel, J. Rodriguez-Carvajal, in: Proceedings of the European Powder Diffraction Conference (EPDIC7), Mater. Sci. Forum, (2000), Book of Abstracts, p. 118.
- [58] E. Parthe, L. Gelato, B. Chabot, M. Penzo, K. Cenozual, R. Gladyshevskii, TYPiX Standardized Data and Crystal Chemical Characterization of Inorganic Structure Types, Springer-Verlag, Berlin, Heidelberg, 1994.
- [59] M.V. Bulanova, P.N. Zheltov, K.A. Meleshevich, P.A. Saltykov, G. Effenberg, J. Alloys Compd. 345 (2002) 110.
- [60] P. Schobinger-Papamantellos, K.H.J. Buschow, J. Alloys Compd. 198 (1993) 47.
- [61] A. Palenzona, F. Canepa, P. Manfrinetti, J. Alloys Compd. 194 (1) (1993) 63.
- [62] M.E. Schlesinger, J. Phase Equilib. 13 (1) (1992) 54.
- [63] A.V. Morozkin, A.E. Bogdanov, R. Welter, J. Alloys Compd. 340 (1–2) (2002) 49.
- [64] A.V. Gribanov, et al., 2010, in preparation.
- [65] A.C. Lawson, J.L. Smith, J.O. Willis, J.A. O'Rourke, J. Faber, R.L. Hitterman, J. Less-Common Met. 107 (2) (1985) 243.
- [66] R.A. Gordon, F.J. DiSalvo, Z. Naturforsch. B 51 (1) (1996) 52.
- [67] T.B. Massalski, H. Okamoto, P.R. Subramanian, L. Kacprzak, Binary Alloy Phase Diagrams, ASM International, Materials Park, OH, USA, 1990.
- [68] P. Villars, L.D. Calvert (Eds.), Pearson's Handbook of Crystallographic Data for Intermetallic Phases, ASM, Materials Park, OH, USA, 1991.
- [69] O. Sologub, P. Salamakha, G. Bocelli, C. Godart, T. Takabatake, J. Alloys Compd. 312 (1–2) (2000) 172.
- [70] A. Raman, J. Less-Common Met. 26 (2) (1972) 199.
- [71] G.L. Olcese, J. Less-Common Met. 33 (1) (1973) 71.
- [72] J. Le Roy, J.-M. Moreau, D. Paccard, E. Parthe, Acta Crystallogr. B 33 (8) (1977) 2414.
- [73] A. Raman, J. Less-Common Met. 48 (1) (1976) 111.
- [74] A.E. Dwight, R.A. Conner Jr., J.W. Downey, Acta Crystallogr. 18 (5) (1965) 835.
- [75] F. Canepa, M. Minguzzi, G.L. Olcese, J. Magn. Mater. 63–64 (1987) 591.
- [76] H. Ghassem, A. Raman, Z. Metallkd. 64 (3) (1973) 197.
- [77] E.V. Sampathkumaran, L.C. Gupta, R. Vijayaraghavan, J. Magn. Mater. 31–34 (1) (1983) 413.
- [78] T. Mihalisin, A. Harrus, S. Raaen, R.D. Parks, J. Appl. Phys. 55 (6) (1984) 1966.
- [79] I.R. Harris, M. Norman, W.E. Gardner, J. Less-Common Met. 29 (3) (1972) 299.
- [80] S.K. Malik, S.K. Dhar, R. Vijayaraghavan, Pramana 22 (3–4) (1984) 329.
- [81] L. Schellenberg, J.L. Jorda, J. Muller, J. Less-Common Met. 109 (2) (1985) 261.
- [82] I. Engström, Acta Chem. Scand. 19 (1965) 1924.
- [83] Y. Muro, D. Eom, N. Takeda, M. Ishikawa, J. Phys. Soc. Jpn. 67 (10) (1998) 3601.
- [84] M.B.T. Tchokonte, P.D. du Plessis, A.M. Strydom, Solid State Commun. 117 (5) (2001) 321.
- [85] A. Krimmel, M. Reilmis, A. Loidl, Appl. Phys. A 74 (2002) S695.
- [86] C. Godart, C.V. Tomy, L.C. Gupta, R. Vijayaraghavan, Solid State Commun. 67 (7) (1988) 677.
- [87] B. Chevalier, J. Etourneau, J. Rossat Mignod, R. Calemczuk, E. Bonjour, J. Phys.: Condens. Matter 3 (12) (1991) 1847.
- [88] A. Gribanov, A. Yatsenko, Yu. Seropegin, J. Kurenbaeva, N. Kocherov, O. Bodak, Visnyk Lviv Univ. Ser. Chim. 39 (2000) 96.